Access to 2,6-Disubstituted Piperidines: Control of the Diastereoselectivity, Scope, and Limitations. Applications to the Stereoselective Synthesis of (−)-Solenopsine A and Alkaloid (+)-241D

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S Supporting Information

[AB](#page-14-0)STRACT: [Scope and lim](#page-14-0)itations in the diastereoselective preparation of 2,6-cis or 2,6-trans disubstituted piperidines are described, through intramolecular reaction of chiral β′ carbamate- α , β -unsaturated ketone. This methodology has been applied to the total synthesis of a few well chosen examples, such as (−)-solenopsine A and alkaloid (+)-241D.

ENTRODUCTION

Substituted piperidines and their analogues are key structural units in numerous naturally occurring alkaloids and in a number of successful pharmaceutical compounds¹ For this reason, a number of methodologies for the elaboration of these structures have been described^{2−5} especially when [st](#page-14-0)ereogenic centers are involved. In particular, those possessing a chiral center at C-2 and/or C-6, ster[eose](#page-14-0)lectivity that is essential for the defined activity, have attracted much attention because they are one of the most common framework encountered in many interesting compounds that exhibit a broad range of biological activities. For example, (−)-solenopsin A and (−)-isosolenopsin A (active components of fire ant venom) are reported to possess a broad range of activities,⁶ alkaloid $(+)$ -241D (isolated from methanolic skin extracts of Panamanian poison frogs Dendrobates speciosus) is active on nicotinic acetylcholine receptors,⁷ and $(-)$ -lasubine II (extracted from plants of the Lythraceae family) has showed cytotoxic, hemolytic, necrotic, insecticid[al](#page-14-0), antibacterial, antifungal, and anti-HIV properties⁸ (Scheme 1). So, developing approaches to allow the stereoselective synthesis of 2,6-dialkylpiperidines is of great value.

For th[is](#page-1-0) purpose, many synthetic methods have been developed including Mannich-type reactions⁹ or ring-closing metathesis.¹⁰ In order to control the diastereoselectivity excess on the positions α and α' of the piperidine c[ore](#page-14-0), some of those routes hav[e f](#page-14-0)ocused on the construction of the ring by C−N ring-closure bond formation,^{11–13} including reductive amination, 14 intramolecular substitution, 15 cyclization of sulfinimides on propargylic ether,¹⁶ intra[mo](#page-14-0)l[ec](#page-14-0)ular allylic substitution with 1,3-[chi](#page-14-0)rality transfer,¹⁷ iminium [io](#page-14-0)n cyclization,¹⁸ $[4 + 2]$ cycloaddition of ald[im](#page-14-0)ines, 19 intramolecular aza- $[2,3]$ -Wittig r earrangement,²⁰ c[ata](#page-14-0)lyzed hydroamination²¹ [or](#page-14-0) Michael

addition. $13,22$ Therefore, all of these methods show that there is always a considerable interest in developing stereoselective access t[o 2,6](#page-14-0)-dialkylpiperidines. However, even though cis-2,6 disubstituted piperidines are readily accessible, only a few methods have been devoted to the synthesis of trans-2,6 disubstituted isomers.23−³⁰

During the course of our recent studies on the asymmetric synthesis of 2,6-dis[ub](#page-14-0)[stit](#page-15-0)uted piperidines by C−N bond formation, we have demonstrated that the Michael-type cyclization,³¹ using $β'$ -carbamate- $α, β$ -unsaturated ketone 1 as key precursor, induced systematically and predominantly the formation [of](#page-15-0) a piperidine ring with the 2,6-trans configuration (Scheme 2). The relative stereochemistry was confirmed by further transformation of the trans derivative in known chiral compoun[d](#page-1-0) 3^{30} with an 95% ee.

In order to establish this new approach as a general method for the prepa[rat](#page-15-0)ion of chiral 2,6-disubstituted piperidines and to understand the requirements for the best selectivity, we have synthesized various $β'$ -carbamate-α, $β$ -unsaturated ketones and tested their cyclization reaction using different conditions.

■ RESULTS AND DISCUSSION

General Synthesis of a Wide Range of β' -Carbamate- α , β -unsaturated Ketones. We have previously shown that the necessary $β'$ -carbamate-α, $β$ -unsaturated ketone 1 could be easily obtained from the corresponding α , β -unsaturated methylester in 6 steps with an overall yield of about $30\%^{31}$ (Scheme 3).

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Scheme 1

 a (a) Davies amine, BuLi, THF, −78 °C; (b) H₂, Pd(OH)₂/C, MeOH; (c) Na₂CO₃, CbzCl, CH₂Cl₂/H₂O; (d) NaOH 1 N, MeOH; (e) CDI, (MeO)MeNH.HCl; (f) Mg, 1-bromo-2-propene, THF, 0 °C.

de > 95%

Scheme 4^a

0	R ₂ O	R ₁ O	R ₂ O																										
4a-d	6a-b	R ₁ OCH ₃	R ₁ OCH ₂	R ₂ OCH ₃	R ₁ O	R ₂ OCH ₃	R ₁ OCH ₂	R ₁ OCH ₂	R ₂ OCH ₃	R ₃ OCH ₃	R ₁ OCH ₂	R ₂ OCH ₃	R ₃ OCH ₃	R ₃ OCH ₃	R ₄ OCH ₂	R ₅ OCH ₂	R ₆ OCH ₂	R ₇ OCH ₂	R ₈ OCH ₂	R ₉ OCH ₃	R ₁ OCH ₂ OCH ₂	R ₁ OCH ₂ OCH ₂	R ₂ ECH $R_3 = ph(p-NO_2)$	886 % % K ₁ OCH ₂ ECH $R_3 = ph(p-NO_2)$	91 % % K ₁ OCH ₂ OCH ₃	79 % % K ₂ ECH $R_3 = ph(p-NO_2)$	91 % % K ₃ A H ₄ OCH ₂	91 % % K ₄ OCH ₄ OCH ₄	R ₅ OCH ₄ OCH ₅ OCH ₅ OCH ₆ </td

 a (a) Davies amine, BuLi, THF, −78 °C; (b) H₂, Pd(OH)₂/C, MeOH; (c) Na₂CO₃, R²CO₂Cl, CH₂Cl₂/H₂O; (d) BuLi, (EtO)₂P(O)Me, THF, −78 $\rm ^{\circ}C;$ (e) Ba(OH)₂, THF/H₂O (40:1), R³CHO. In the case of enone 7**k**, we obtained a mixture of Z and E stereoisomers in a 60:40 ratio, respectively.

As Grignard's reagents do not allow the use of a wide range of functionalities, we have devised a general and simple method to access a variety of compounds of type 1 easily by using a more convenient way through a Wittig-Horner-Emmons³² reaction as the key step (Scheme 4). By this method, t[he](#page-15-0)

needed compounds were prepared in four steps from the corresponding α , β -unsaturated methylester according to described procedure.³³

 $E/Z: 3/1$

Addition of enantiopure lithium N-benzyl-N-α-methylbenzylamide on α , β -unsat[ura](#page-15-0)ted ester 4 following by hydrogenation

to the corresponding primary amine and further protection as a carbamate gave the β -amino methylester 5a−g. After purification, the ester function was transformed into the ketophosphonate 6a−g by treatment with 2.5 equiv of diethyl lithiomethylphosphonate³⁴ in THF at -78 °C, in moderate yields. Over the years, many examples of base-promoted Wittig−Horner−Emmo[ns](#page-15-0) reaction have been reported in scientific literature,^{35−37} and various combinations of bases and solvents $(K_2CO_3/CH_3CN, DBU/THF, NaH/THF, Et_3N/$ LiCl/CH₃CN or Ba $(OH)_{2}/(THF/H_{2}O)$, etc.) have been used. In our case, we have found that the use of 1.3 equiv of $Ba(OH)$ ₂ in biphasic medium THF/H₂O (40:1) was the more general and convenient route to obtain compounds 7a−z with good to excellent yield.

Scope and Limitations of the Michael-Type Cyclization. As mentioned previously, 31 we have shown that compound 1 could be easily transformed diastereoselectively by intramolecular Michael-type rea[ctio](#page-15-0)n in 2,6-disubstituted-Nprotected-4-ketal piperidine $(2a/2b)$ as a mixture of *cis/trans* isomers in which the trans conformation represents the major compound (Scheme 2). We have also shown that the character Z or E of the geometry of the double bond in compound 1 did not have any influ[en](#page-1-0)ce on the diastereoselectivity of the cyclization reaction, as similar results have been obtained starting from either stereoisomer (E) or (Z) of 1 treated in the same optimized conditions $(0.2 \text{ equiv of } p\text{-tolueness.})$ acid monohydrate, 5 equiv of ethylene glycol, 5 equiv of trimethyl orthoformate, which has been used here as solvent and as a water scavenger).

So, we decided to use this protocol for the cyclization of a range of dissymmetric (aliphatic/aromatic for R^1 and $\text{R}^3)$ enecarbamates of type 7, hoping to evaluate at first the influence of steric hindrance on the selectivity. For a better evaluation of the *cis/trans* ratio (¹H NMR) the mixture of ketals 8 and 9 were directly converted into the more stable thioketals 10 and 11 by known procedure, using 1,2-ethane dithiol in the presence of boron trifluoride diethyl etherate, as it has been shown that this transformation induced no variation of the diastereoisomeric ratio (Scheme 5 and Table 1).

According to Table 1, for the defined conditions, the selectivity observed for the cyclization reaction is predominantly in favor of the trans isomer, which is the less stable conformation for a 2,6-disustituted piperidine. This de is markedly dependent on a lot of factors, namely, the nature of the nitrogen protective group and also the nature of steric hindrance $(R^1, R^2,$ and $R^3)$ on compounds 7. On the one hand, when R^1 and R^3 are fixed $(R^1 = Ph, R^3 = Me)$; entries 1, 2, and 3), a strong steric hindrance around the nitrogen atom is necessary to induce a good diastereoselectivity. On the other hand, when R^1 = Ph and R^3 is a much longer alkyl chain (propyl or nonyl; entry 7, 8, and 9), the ethyl carbamate function is sufficient to ensure predominantly the formation of the trans isomer, however with a small diastereoselectivity

Table 1. Scope of Intramolecular Michael Reaction

a Diastereoisomeric ratio is determined by integration of characteristic protons of the piperidine ring in the crude $H¹ NMR$ spectra (*trans* and cis isomers were determined according to their respective coupling
constants). ^bIsolated yield of pure diastereoisomeric mixture after chromatography on silica gel. ^c10a/11a and 10d/11d are enantiomers; for 10a/11a (10d/11d) the benzyl carbamate was cleaved during the thioketalation process, leading to the free amine on the piperidine ring, see Experimental Section).

exc[ess \(de = 24%\). Fu](#page-7-0)rthermore, changing ethylene glycol to 1,3-propane diol increases the de up to 90%, showing by this way the importance of the resulting keto protecting group. Permutation of R^1 and R^3 (entry 3 versus entry 5) does not show a significant influence on the diastereoselectivity unless the alkyl chain is much longer than a methyl group (entries 5, 6, and 10), or if $R¹$ is a phenyl group (entry 11), in which the observed de is again around 90%. Another observation is the dependency of the cyclization on the nature of the alcohol used. So, when R^1 = Me and R^3 = pyridine (7i), the formation of the

Scheme 6

piperidine ring is not observed if ethylene glycol is used to form the ketal (entry 12). The only product that can be identified (by H NMR spectroscopy) is the ketal 12, in which the double bond is exclusively in a E conformation $(J = 15.5 \text{ Hz})$. Assuming that the relative hindrance between the dioxolane group and the pyridine was too high, we decided to realize the reaction with a less crowded acetal. To our delight, when compound 7i was engaged (entry 13) in the presence of trimethyl orthoformate and p-toluenesulphonic acid, leading to the in situ formation of methanol, we could now isolate the corresponding piperidines (10k/11k) with a de of 70% in favor of the trans conformation (Scheme 6).

This last result confirmed the fact that the first step, for the elaboration of the piperidine ring, is the formation of the ketal on compounds 7 and thus can constitute the critical step for the diastereoselectivity of the reaction. Moreover, the formation of the ketal induces the E or Z configuration of the double bond, which is dependent on the nature of the ketal. This hypothesis based on the role of the geometry of the double bond of a Michael acceptor in the control of the diastereoselectivity, during the formation of 2,6-disubstituted piperidines, has already been put forward by Banwell and co-workers,³⁸ although in the case of an exocyclization process. They had demonstrated that the geometry of the double bond conduct[ed](#page-15-0) to two different transition states, resulting in the formation of 2,6-cis or 2,6-trans piperidine. Thus, reducing this fact to our model, we suppose that the formation of the 2,6-trans

piperidine or 2,6-cis piperidine can be correlated with the geometry of the double bond of the crucial intermediate acetal.

In order to confirm this hypothesis we engaged the compound 7j $(Z/E \text{ conformers} = 40/60)$ in reaction with trimethyl orthoformate, with and without ethylene glycol and in the presence of acid (Scheme 7). Compound 7j was specially chosen for the strong steric hindrance that could be generated in the transition state. At this stage, no formation of a piperidine ring was expected, but rather the formation of the intermediate ketal form and the possibility to measure the corresponding coupling constant of both intermediates, to validate our hypothesis.

Then, it was possible to identify two α , β -unsaturated ketals $7ka(E)$ and $7kb(Z)$, a dioxolane and a dimethylketal, respectively. The coupling constant values for the double bond in ¹H NMR spectroscopy, demonstrated the existence of two different stereoisomers depending on the ketal formed. The E configuration (J_{HAHB} = 15.5 Hz) was observed for the cyclic ketal 7 $\text{ka}(E)$, whereas the Z (J_{HAHB} = 8.4 Hz) was observed for the dimethyl ketal $7kb(Z)$. As for compound $7ka(E)$, if trimethyl orthoformate is added to the mixture, a transacetalation reaction is observed, leading to the formation of $7kb(Z)$. Thus, this result first of all confirmed the formation of the ketal as the first step of the transformation and, second, strongly suggested that the diastereoselectivity of the piperidine formed can be dependent on the configuration of the double bond of the intermediate ketal in the transition state. However

 a Diastereoisomeric ratio is determined by integration of characteristic protons of the piperidine ring in the crude H 1 NMR spectra (*trans* and *cis* Exacted constants that to determined by misglation of characteristic process of the paperial ing in the cause 11 Tank opening (mine that the on silica gel.

at this stage we cannot connect the configuration of the double bond with the configuration of the piperidine formed. We can only assume the existence of two different transition states corresponding to the formation of two different acetals.

Therefore, to confirm the real role of this geometry on the resulting diastereoisomeric excess for the piperidine generated, we selected the compound 7q, which gave a de \geq 90% in favor of the trans isomer (Table 1, entry 6) and compared the importance of the nature of the alcohol on the result when methanol, ethylene glycol, o[r](#page-2-0) propan-1,3-diol is used in the cyclization process (Table 2); both enantiomers of 7q were tested. The quantity of acid was fixed at 0.2 equiv, and all the crude mixtures 8f,9f were directly converted into, respectively, the corresponding 4-thioketal piperidine 10f and 11f (Table 2). The diastereomeric excess was as before, calculated according to the ¹H NMR values.

Cyclization was observed in all cases with a good overall yield, but a significant difference in the de was observed. The higher diastereoselectivity was obtained (Table 2, entry 1 and 2) when ethylene glycol is used to form the ketal, corresponding to the more overcrowded intermediate. On the contrary, lower diastereoselectivity is observed when methanol is used (Table 2, entry 4), as the dimethyl acetal gave a higher flexibility to the intermediate. Both enantiomers of 7q gave the same result (Table 2, entries 1 and 2). If steric hindrance appeared here as the predominant factor for the stereoselectivity of the reaction, however, in all cases the trans isomer was obtained. So, in order to reinforce the existence of two different transition state according to the geometry of the double bond of the ketal, we envisaged that this critical step could be under a kinetic control. Compound 7q was now engaged under two different experimental protocols: on the one hand with ethylene glycol and on the other hand with methanol, and for each case increasing quantities of acid, from catalytic to stoichiometric, were used (Table 3).

When ethylene glycol is used for the formation of the ketal, the quantity of acid does not affect the diastereoisomeric excess obtained for 10f/11f (entries 1−4), and only the reaction time is reduced. After 24 h (entry 5), an epimerization through retro-Mannich or retro-Michael reaction is observed, the more stable *cis* isomer 11f becoming now the major compound. On the contrary, when methanol is used (entries 6−10) to generate the ketal, there is a significant difference in the diastereoiso-

Table 3. Kinetic Effect on the Stereoselectivity in the Cyclization Process

entry	alcohol (equiv)	p -TsOH/ H ₂ O (equiv)	cyclization reaction time $8f/9f$ (min)	10f/11f $(\%)^a$
1	HO - (CH_2) ,- $OH (5)$	0.1	25	85/15
\mathfrak{p}	HO - (CH_2) ,- $OH (5)$	0.2	20	85/15
3	HO - (CH_2) ,- $OH (5)$	0.5	15	86/14
4	HO - (CH_2) ,- $OH (5)$	1	10	87/13
5	HO - (CH_2) ,- $OH (5)$	0.2	24 h	40/60
6	MeOH(1)	0.02	45	50/50
7	MeOH(1)	0.05	40	55/45
8	MeOH(1)	0.1	20	64/36
9	$MeOH$ (0 or 1)	0.2	15	69/31
10	MeOH $(0)^b$		10	72/28

a Diastereoisomeric ratio is determined by integration of characteristic protons of the piperidine ring in the crude $H¹ NMR$ spectra (*trans* and cis isomers were determined according to their respective coupling constants) ^bWhen the concentration of acid is more than 0.2 equiv, the degradation of trimethyl orthoformate is sufficient to generate methanol in situ.

meric excess depending on the quantity of catalyst used, and this evolution is in agreement with a kinetic effect.

To evaluate the importance of this kinetic effect in the transition state, compared to the steric effect, we substituted the aromatic ring of 7a and 7q (respectively table 3, entry 5, 39% of *trans* isomer and entry 6 , \geq 95% of *trans* isomer, with the use of ethylene glycol and 0.2% acid) with various ERG or EWG groups in the ortho, meta, or para position of the aromatic ring, and we used the methanol generated by the trimethyl orthoformate for the formation of the ketal. Results on the selectivity so obtained are reported in Table 4.

Strong EWG located in para position on the aromatic ring for 7r and 7d (Table 4, entries 1 and 2) or placed on a conjugated system 7j (Table 4, entry 9) led predominantly and respectively to 83%, 89[%,](#page-5-0) or 90% in favor to the trans isomer, but no significant influence [on](#page-5-0) the selectivity can be related to the quantity of acid used. When the EWG is in the meta position, (compound 7c, Table 4, entry 3), a deactivating position, the opposite diastereoselectivity (40/60) is observed and a longer reaction time is req[ui](#page-5-0)red. Concerning the ortho position (compound 7b, Table 4, entry 4), the EWG effect is

Table 4. Electronic Effects

			O	С				
		$(CH_3O)_3CH$ (5 equiv.)	r_{R^3} R^2	R^2 R^3				
R^3 p-TsOH (0.2-1 equiv.) R^2								
	$\overline{7}$	r. t.	8 Trans	9 Cis				
	Ketone 7	Reaction time (min)		(Trans/Cis)				
Entry		p -TsOH/H ₂ O (x equiv.)		8/9 $(\%)^{a}$ [yield $\%$] ^b				
		(90min)	(20min)	83/17 [71]				
$\mathbf{1}$	$\frac{1}{2}$	$(0.2$ equiv.)	$(1$ equiv.)	(81/91)				
		(120min)	(30 min)	89/11 [80]				
$\overline{2}$	$\sqrt{2}$ 7d	$(0.2$ equiv.)	$(1$ equiv.)	(8m/9m)				
		(60min)		40/60 [71]				
3	7c	$(1$ equiv.)		(8n/9n)				
			(60min)	60/40 [67]				
$\overline{\mathbf{4}}$	7 _b	$(1$ equiv.)		(80/90)				
		(45 min)		33/67 [73]				
5	7g	$(1$ equiv.)		(8p/9p)				
	(60min) 7f $(1$ equiv.)			29/71 [75]				
6				(8q/9q)				
		(180min)		56/43 [76]				
$\overline{7}$	7 _h $(1$ equiv.)			(8r/9r)				
				decomposition				
8	же $7\mathrm{e}$							
	(240min)			90/10 [93]				
9	" 7j	$(0.2$ equiv.)		(8s/9s)				

 a Diastereoisomeric ratio is determined by integration of characteristic protons of the piperidine ring in the crude H^1 NMR spectra (trans and cis isomers were determined according to their respective coupling constants). ^b Isolated yield of pure diastereoisomeric mixture after chromatography on silica gel.

counterbalanced by steric hindrance in the transition state, and therefore a small diastereoisomeric excess (60/40) is obtained. As expected, the presence of an ERG group in the ortho or para position (Table 4, entries 5 and 6) led predominantly to the cis derivative (30/70). However, if this effect is too strong, decomposition of the starting material is observed (Table 4, entry 8). Thus, when compounds 7 have an aromatic or a conjugated system as a substituent on the double bond, a lot of parameters (steric hindrance, angle pressure, kinetic effect and electronic effect) has to be considered to access a high selectivity, and this selectivity is in favor of the trans isomer of the piperidine.

So, in order to validate the existence of two transition states, according to the alcohol used in the cyclization step, we carried out the reaction with four representative ketones 7k,l,t,u in which only a steric hindrance was induced by the size of an alkyl chain. As usual, we measured the outcomes observed for the diastereoselectivity when a cyclic ketal or a dimethyl ketal formation was involved. Here too, we converted all the crude mixture 8 and 9 or 8′ and 9′ directly into the corresponding 4 thioketal piperidines 10 and 11 (Table 5).

According to the results obtained (Table 5) and the stereoselectivity observed, it becomes [ev](#page-6-0)ident now to affirm the formation of two different transition states in t[he](#page-6-0) formation of the piperidine ring. Both transition states are strongly dependent on the alcohol used to generate the ketal on the α , β unsaturated ketone. As we have mentioned in Table 3, there are two behaviors according the nature of the ketal group. When dioxolane is used, the trans isomer is the major prod[uc](#page-4-0)t formed, at ∼80%, with a catalytic amount of acid (Table 5, entries 1, 4, 5, 7). However when a stoichiometric quantity of acid is used, a fast epimerization process occurs (retro Manni[ch](#page-6-0) or Michael reaction), and consequently a higher formation of the thermodynamic specie, namely, the 2,6-cis piperidine, is observed, even if the trans isomer was the first formed. In contrast, as observed in table 3, there is no kinetic effect on the cyclization when methanol is used. In this case, the cis isomer was the major product, ∼8[0%](#page-4-0) of the reaction; whatever the quantity of acid engaged (Table 5, entries 3, 6, 8), the

Table 5. Michael Cyclization of Enones 7 Bearing Aliphatic Substituents

 a Diastereoisomeric ratio is determined by integration of characteristic protons of the piperidine ring in the crude $\rm H^1$ NMR spectra (trans and cis isomers were determined according to their respective coupling constants). ^BIsolated yield of pure compounds after chromatography on silica gel.
CMeOH was generated in situ by decomposition of trimethyl orthoformate MeOH was generated in situ by decomposition of trimethyl orthoformate.

Scheme 8^a

 a ^a(a) OH(CH₂)₂OH, (CH₃O)₃CH, p-TsOH; (b) SH(CH₂)₂SH, BF₃·Et₂O, CH₂Cl₂; (c) (Boc)₂O, DMAP, CH₂Cl₂; (d) W-2 Raney nickel, EtOH, reflux; (e) TFA, $CH₂Cl₂$.

thermodynamic product was formed predominantly. Then, as for Banwell's studies, our results are in agreement with the formation of two possible conformations depending of the conformation of the double bond before nucleophilic attack of the carbamate, leading to the formation of the piperidine. At this stage, after examination of all potential parameters that can

interfere with diastereoselectivity, it becomes easy to prepare either the trans or the cis isomer by a wise choice of experimental conditions. In order to demonstrate this, we applied these protocols to the synthesis of (−)-solenopsin- $A^{39,40}$ and $(+)$ -alkaloid 241D⁴¹⁻⁴³ (Schemes 8 and 9). (−[\)-So](#page-15-0)lenopsin A can be rapi[dly](#page-15-0) [pr](#page-15-0)epared from compound

Scheme 9^a

 a^a (a) (CH₃O)₃CH, pTsOH; (b) TFA/H₂O, CH₂Cl₂; (c) Pd/C (5%), MeOH, H₂, 1 atm; (d) NaBH₄, MeOH.

7p by using for the cyclization step ethylene glycol, which led to the trans isomer 8w as a major product (82%, Scheme 8). After subsequent transformations that we have already described in a previous paper⁴⁴ (thioketalation 10w, Boc protectio[n,](#page-6-0) desulfuration 12), (−)-solenopsin A was obtained in 10 steps, after regeneration o[f th](#page-15-0)e free amine, from the α , β -ethylenic ester 4, with an overall yield of 9% with $[\alpha]_D = -1.21$ (c 0.94 MeOH, $\text{lit.}^{34}[\alpha]_{\text{D}} = -1.30$ (c 1.30 MeOH).

To reach (+)-alkaloid 241D, methanol was used for the cy[cli](#page-15-0)zation, starting from compound 7o. The intermediates 8′x/ 9′x were obtained in a ratio of 15/85 in a favor of the cis isomer. After separation on deactivated silica, 9′x was ketodeprotected using a 40% aqueous trifluoroacetic acid solution at room temperature to give the corresponding piperidones 13 in very good yield. N-Deprotection of the piperidone 13, followed by reduction with $NaBH_4$ gave selectively $(+)$ -alkaloid 241D in 9 steps from the α , β -ethylenic ester 4 with an overall yield of 16% with a de \geq 95% and an ee of 92% (Scheme 9).

■ CONCLUSION

In conclusion, we have described herein a methodology to prepare stereoselectively either 2,6-cis or 2,6-trans disubstituted piperidines. The efficient of this methodology has been demonstrated through the asymmetric synthesis of (−)-solenopsine A and (+)-alkaloid 241D together with their respective isomer in C-6, demonstrating in this way that this strategy will be applied efficiently to the total synthesis of other piperidinic alkaloids exhibiting important biological interest.

EXPERIMENTAL SECTION

General. Organic solutions were dried over $Na₂SO₄$ and filtered. When anhydrous solvents were used, they were prepared as follows: tetrahydrofuran (THF) was distilled under N_2 from sodium benzophenone ketyl and used immediately; anhydrous acetonitrile was freshly distilled from CaH_2 . All ¹H NMR and ¹³C spectra were measured in CDCl₃ or C_6D_6 and recorded on a Brüker 400 MHz (101) MHz for ¹³C) spectrometer using TMS as the internal standard. Chemical shifts are expressed in ppm and J values are given in hertz. The following abbreviations are used: singlet (s), broad singlet (brs), doublet (d), doubled doublet (dd), triplet (t), multiplet (m). High resolution mass spectroscopy (HRMS, TOF) were carried out in electrospray mode. Monitoring of the reactions was performed using silica gel TLC plates. Spots were visualized by UV light at 254 nm. Flash chromatography columns were performed using silica gel 60 (70−230 mesh).

General Procedure for the Synthesis of β -Aminoesters 5. (R)-Methyl-3-(ethoxycarbonylamino)butanoate 5a. To a cold solution (0 °C) of (+)-(R)-N-benzyl-N- α -methyl benzylamine (23.0 mL, 110 mmol, 1.1 equiv) in dry THF (280 mL) was added slowly under argon n-butyl lithium (75.0 mL, 1.6 M in hexane, 120 mmol, 1.2 equiv). The resultant pink solution of lithium amide was stirred for 30 min at 0 °C and then cooled to −78 °C before dropwise addition of a solution of methyl crotonate (10.0 mL, 100 mmol, 1 equiv) in dry THF (100 mL). The mixture was stirred at −78 °C for 3.5 h. Then, a saturated aqueous solution of NH4Cl (100 mL) was added slowly, and the

resulting solution was allowed to warm to room temperature. The solution was extracted twice with ethyl acetate. Combined organic extracts were dried over Na_2SO_4 , filtered, and evaporated. The crude product was added to a suspension of 10% Pd/C (5.00 g) in methanol (200 mL). The mixture was placed on a Parr apparatus and stirred under a hydrogen atmosphere (60 psi) for 4 days. The catalyst was then removed by filtration on Celite. The residue was concentrated in vacuo and dissolved in dichloromethane (200 mL) and water (200 mL). Then, sodium carbonate (42.4 g, 400 mmol, 4.0 equiv) and ethyl chloroformate (28.5 mL, 200 mmol, 2 equiv) were added dropwise. The resulting solution was stirred at room temperature for 3 h. The aqueous material was extracted with dichloromethane and the combined organic extracts were dried over $Na₂SO₄$, filtered, and concentrated in vacuo. Purification by chromatography on silica gel (cyclohexane/EtOAc 9:1 to 5:5) afforded 5a as a yellow oil (10.8 g, 57% over 3 steps): $[\alpha]_{D} = -35.60$ (c 0.99, CHCl₃), lit.²⁷ $[\alpha]_{D} =$ -37.07 (c 1.00, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 5.03 (brs, 1H, N[H\)](#page-15-0), 4.03 (m, 3H), 3.62 (s, 3H), 2.46 (d, J = 6.9 Hz, 2H), 1.16 (t, $J = 6.9$ Hz, 3H), 1.15 (d, $J = 6.6$ Hz, 3H). Spectral data are identical with those reported. 27

(R)-Methyl-3-(ethoxycarbonylamino)butanoate 5a. (Starting from 0.100 mol of [4a](#page-15-0)) Yellow oil, 10.8 g, yield = 57 %) Spectral data are identical with those reported:⁴⁵ $[\alpha]_{\text{D}} = -37.07$ (c 1, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 5.03 (brs, 1H), 4.03 (m, 3H), 3.62 (s, 3H), 2.46 (d, $J = 6.9$ $J = 6.9$ Hz, 2H), 1.16 (t, $J = 6.9$ Hz, 3H), 1.15 (d, $J = 6.6$ Hz, 3H).

(R)-Methyl-3-(benzyloxycarbonylamino)butanoate 5b. (Starting from 0.100 mol of $4a$) Yellow oil, 15.6 g, yield = 62%. Spectral data are identical with those reported:⁴⁶ $\left[\alpha\right]_D$ = +16.9 (*c* 1.4, CHCl₃). ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 5.03 (brs, 1H), 4.03 (m, 3H), 3.62 (s, 3H), 2.46 $(d, J = 6.9 \text{ Hz}, 2\text{H}), 1.16 \text{ } (t, J = 6.9 \text{ Hz}, 3\text{H}), 1.15 \text{ } (d, J = 6.6, 3\text{H}).$ $(d, J = 6.9 \text{ Hz}, 2\text{H}), 1.16 \text{ } (t, J = 6.9 \text{ Hz}, 3\text{H}), 1.15 \text{ } (d, J = 6.6, 3\text{H}).$ $(d, J = 6.9 \text{ Hz}, 2\text{H}), 1.16 \text{ } (t, J = 6.9 \text{ Hz}, 3\text{H}), 1.15 \text{ } (d, J = 6.6, 3\text{H}).$

(R)-Methyl-3-(benzyloxycarbonylamino)butanoate 5c and (S)- Methyl-3-(benzyloxycarbonylamino)butanoate 5c'. (Starting from 0.100 mol of 4b) Yellow oil, 16.7 g, yield = 77%. R enantiomer: $[\alpha]_D$ = +41.5 (c 1.03, CHCl₃), S enantiomer: $[\alpha]_D = -40.9$ (c 1.035, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 4.98 (brs, 1H), 4.02 (m, 2H), 3.90 (m, 1H), 3.61 (s, 3H), 2.49 (dd, J = 15.8, 4.8 Hz, 1H), 2.43 (dd, J = 15.8, 5.3 Hz, 1H), 1.48–1.22 (m, 2H), 1.16 (t, $J = 7.0$ Hz, 3H), 0.85 (t, $J =$ 7.2 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 172,0, 156.0, 60.7, 51.6, 47.7, 38.9, 36.6, 19.3, 14.6, 13.8. HRMS-ESI (M + Na) m/z calcd for C10H19NO4Na 240.1212, found 240.1216.

 (R) -Methyl-3-(ethoxycarbonylamino)undecanoate 5d. (Starting from 0.100 mol of 4c) Yellow oil, 20.9 g, yield = 73%. $[\alpha]_D$ = +29.2 (c 1.05, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.07 (brs, 1H), 4.22– 4.03 (m, 2H), 3.98 (m, 1H), 3.69 (s, 3H), 2.57 (dd, J = 15.1, 4.6 Hz, 1H), 2.51 (dd, J = 15.1, 5.1, Hz 1H), 1.55−1.45 (m, 2H), 1.41−1.10 (m, 15H), 0.89 (t, J = 6.9 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 172.1, 156.1, 60.7, 51.5, 48.0, 38.9, 34.4, 31.8, 29.4, 29.3, 29.2, 26.1, 22.6, 14.6, 14.1; HRMS-ESI (M + Na) m/z calcd for $C_{15}H_{29}NO_4Na$ 310.1994, found 310.1996.

(S)-Methyl-3-(ethoxycarbonylamino)-3-phenylpropanoate 5e. (Starting from 0.100 mol of $4d$) Yellow oil, 18.0 g, yield = 72% . $[\alpha]_{\text{D}} = -9.7$ (c 0.99, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 7.36– 7.25 (m, 5H), 5.75 (brs, 1H), 5.17 (m, 1H), 4.11 (q, J = 7.0 Hz, 2H), 3.62 (s, 3H), 2.91 (dd, J = 15.5, 6,0 Hz, 1H), 2.84 (dd, J = 15.5, 5.9 Hz, 1H), 1.23 (t, J = 7.0, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 171.3, 155.8, 140.9, 128.6, 127.6, 126.2, 61.0, 51.8, 51.7, 40.5, 14.6; HRMS-ESI (M + Na) m/z calcd for $C_{13}H_{17}NO_4$ Na 274.1055, found 274.1069.

(S)-Methyl-3-(tert-butoxycarbonylamino)-3-phenylpropanoate **5f.** (Starting from 0.100 mol of 4d) Yellow oil, 21.2 g, yield = 76%). Spectral data are identical with those reported.⁴

(S)-Methyl-3-(benzyloxycarbonylamino)-3-phenylpropanoate 5g . (Starting from 0.100 mol of 4d) Yellow oil[, 2](#page-15-0)0.7 g, yield = 66%. Spectral data are identical with those reported⁵²: $[\alpha]_D = -16.1$ (c 0.97, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 7.22 (m, 10 H), 5.73 (brs, 1H), 5.09 (m, 1H), 5.01 (d, $J = 12.3$ Hz, 1H), 4.97 (d, $J = 12.3$ Hz, 1H), 3.50 (s, 3H), 2.81 (dd, J = 15.3, 5.0 Hz, 1H), 2.74 (dd, J = 15.3, 5.7 Hz, 1H).

General Procedure for the Synthesis of Ketophosphonates 6. (R)-Ethyl 5-(Diethoxyphosphoryl)-4-oxopentan-2-ylcarbamate 6a. To a solution of diethyl methylphosphonate (5.8 mL, 39.7 mmol, 2.5 equiv) in anhydrous THF (15 mL) at −78 °C was added dropwise n-butyl lithium (24.8 mL, 1.6 M in hexane, 39.7 mmol, 2,5 equiv). After 20 min at -78 °C, a solution of 5a (3 g, 15.9 mmol, 1 equiv) in anhydrous THF (15 mL) was added dropwise. After addition, the temperature of the reaction was kept at −78 °C for 30 min, then allowed to reach 0 $^{\circ}{\rm C}$ in 1 h, quenched with a solution of ammonium chloride, and extracted twice with ethyl acetate. After drying over $Na₂SO₄$ and concentration under vacuum, the crude oil was first distillated at low pressure to remove excess diethyl methylphosphonate, and the residue purified by flash chromatography (eluent, cyclohexane/EtOAc 2:1 to EtOAc) afforded compound 6a as a yellow oil (3.3 g, 68% yield): $[\alpha]_{\text{D}} = +33.60$ (c 1.17, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 5.03 (brs, 1H₁), 4.16−3.94 (m, 7H), 3.08 $(dd, J = 23.0, 14.0 \text{ Hz}, 1H), 2.99 \text{ (dd, } J = 22.6, 14.0 \text{ Hz}, 1H), 2.84 \text{ (dd, }$ J = 17.1, 6.0 Hz, 1H), 2.71 (dd, J = 17.1, 5.7 Hz, 1H), 1.33−1.21 (m, 6H), 1.15−1.20 (m, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 200.6, 155.8, 62.6 (d, $J = 6.6$ Hz), 62.5 (d, $J = 6.5$ Hz), 60.5, 49.6, 43.5, 42.9 $(d, J = 127.4 \text{ Hz})$, 20.7, 16.2, 16.1, 14.6; HRMS-ESI $(M + Na)$ m/z calcd for $C_{12}H_{24}NO_6$ PNa 332.1239, found 332.1239.

(R)-Benzyl-5-(diethoxyphosphoryl)-4-oxopentan-2-ylcarbamate **6b.** (Starting from 16 mmol of 5b) Yellow oil, 3.4 g, yield = 57%; $[\alpha]_D$ $= -26.4$ (c 0.85, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.36–7.22 $(m, 5H)$, 5.26 (brs, 1H), 5.00 (s, 2H), 4.03 (m, 5H), 3.05 (dd, J = 23.2, 13.6 Hz, 1H), 2.96 (dd, J = 22.7, 13.6 Hz, 1H), 2.85 (dd, J = 17.3, 5.8 Hz, 1H), 2.70 (dd, J = 17.3, 5.6 Hz, 1H), 1.26 (t, J = 6.2 Hz, 6H), 1.16 (d, J = 6.7 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 200.7, 155.6, 136.6, 128.5, 128.0, 66.5, 62.7 (d, $J = 6.6$ Hz), 62.6 (d, $J = 6.8$ Hz), 49.5, 43.6, 43.3 (d, J = 129.5 Hz), 20.4, 16.3, 16.2; HRMS-ESI (M + Na) m/z calcd for $C_{17}H_{26}NO_6PN$ a 394.1395, found 394.1395.

(R and S)-Ethyl-1-(diethoxyphosphoryl)-2-oxoheptan-4-ylcarbamate 6c and 6c'. (Starting from 16 mmol of 5c) Yellow oil, 3.4 g, yield = 65%; R enantiomer: $[\alpha]_D$ = +43.09 (c 1.03, CHCl₃), S enantiomer: $[\alpha]_{D} = -42.55$ (c 1.075, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 5.00 (d, J = 9.0 Hz, 1H), 4.15−3.97 (m, 6H), 3.90 (m, 1H), 3.09 (dd, $J = 22.9$, 13.5 Hz, 1H), 2.96 (dd, $J = 22.5$, 13.5 Hz, 1H), 2.80 (dd, J = 17.2, 5.9 Hz, 1H), 2.74 (dd, J = 17.2, 5.3 Hz, 1H), 1.48−1.39 (m, 2H), 1.37−1.29 (m, 2H), 1.27 (dt, J = 7.2, 2.0 Hz, 6H), 1.15 (t, J = 7.1 Hz, 3H), 0.84 (t, $J = 7.2$ Hz); ¹³C NMR (101 MHz, CDCl₃) δ 200.9, 156.2, 62.7 (d, J = 6.5 Hz), 62.6 (d, J = 6.6 Hz), 60.6, 48.2, 47.5, 42.9 (d, J = 126.6 Hz), 36.7, 19.3, 16.3, 16.2, 14.6, 13.8; HRMS-ESI $(M + Na)$ m/z calcd for C₁₄H₂₈NO₆PNa 360.1552, found 360.1562.

(R)-Ethyl-1-(diethoxyphosphoryl)-2-oxododecan-4-ylcarbamate **6d.** (Starting from 16 mmol of **5d**) Yellow oil, 4.2 g, yield = 66%; $[\alpha]_D$ $= +30.06$ (c 1.00, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 5.02 (brs, 1H), 4.17−3.99 (m, 6H), 3.95 (m, 1H), 3.07 (dd, J = 23.1, 13.5 Hz, 1H), 2.98 (dd, J = 22.7, 13.5 Hz, 1H), 2.83 (dd, J = 17.4, 5.8 Hz, 1H), 2.75 (dd, J = 17.4, 5.2 Hz, 1H), 1.51−1.41 (m, 2H), 1.39−1.23 (m, 12H), 1.21 (t, $J = 7.2$ Hz, 6H), 1.16 (t, $J = 7.1$ Hz, 3H), 0.86 (t, $J = 7.2$ Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 200.8, 156.1, 62.9 (d, J = 6.6 Hz), 62.8 (d, J = 6.7 Hz), 60.7, 48.2, 48.1, 42.7 (d, J = 127.1 Hz), 34.4, 31.8, 29.4, 29.3, 29.2, 26.1, 22.6, 16.4, 16.3, 14.6, 14.0; HRMS-ESI (M + Na) m/z calcd for C₁₉H₃₈NO₆PNa 430.2334 found 430.2349.

(S)-Ethyl-4-(diethoxyphosphoryl)-3-oxo-1-phenylbutylcarbamate **6e.** (Starting from 16 mmol of **5e**) Yellow oil, 3.4 g, yield = 58%; $[\alpha]_D$ $= +1.65$ (c 1.09, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 7.29–7.21 $(m, 4H)$, 7.16 $(m, 1H)$, 5.72 $(s, 1H)$, 5.09 $(dd, J = 12.7, 6.7 Hz, 1H)$, 4.11−3.94 (m, 6H), 3.26 (dd, J = 16.9, 7.3 Hz, 1H), 3.04 (dd, J = 23.3, 13.1 Hz, 1H), 2.97 (dd, J = 16.9, 12.7 Hz, 1H), 2.93 (dd, J = 22.9, 13.1

Hz, 1H), 1.22 (td, $J = 7.1$, 1.9 Hz, 6H), 1.13 (t, $J = 6.7$ Hz, 3H); ¹³C NMR (101 MHz, CDCl3) δ 199.9, 155.9, 141.4, 128.6, 127.4, 126.3, 62.8 (d, J = 6.2 Hz), 62.6 (d, J = 6.5 Hz), 60.8, 51.1, 49.4, 43.3 (d, J = 125.5 Hz), 16.2 (d, J = 6.1 Hz), 14.6; HRMS-ESI (M + Na) m/z calcd for $C_{17}H_{26}NO_6$ PNa 394.1395, found 394.1414.

(S)-tert-Butyl-4-(diethoxyphosphoryl)-3-oxo-1-phenylbutylcarba*mate* **6f.** (Starting from 16 mmol of **5f**) Yellow oil, 3.9 g, yield = 62% ; $[\alpha]_{\text{D}} = +3.12$ (c 1.01, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.26– 7.12 (m, 4H), 7.16 (m, 1H), 5.44 (brs, 1H), 5.03 (brs, 1H), 4.04−3.95 $(m, 4H)$, 3.20 (dd, J = 16.8, 7.4 Hz, 1H), 3.01 (dd, J = 23.1, 12.9 Hz, 1H), 2.99 (m, 1H), 2.92 (dd, J = 22.7, 12.9 Hz, 1H), 1.32 (s, 9H), 1.21 (td, J = 7.1, 1.9 Hz, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 200.2, 155.1, 133.6, 130.9, 128.5, 126.3, 66.7, 62.8 (d, J = 6.3 Hz), 62.6 (d, J = 6.2 Hz), 50.9, 49.6, 41.0 (d, $J = 127.6$ Hz), 16.3 (d, $J = 5.8$ Hz), 16.2 (d, J = 5.6 Hz); HRMS-ESI (M + Na) m/z calcd for C₁₉H₃₀NO₆PNa 422.1708, found 422.1722.

(S)-Benzyl-4-(diethoxyphosphoryl)-3-oxo-1-phenylbutylcarba*mate* **6g.** (Starting from 16 mmol of 5g) Yellow oil, 4.5 g, yield = 66%; $[\alpha]_D = +8.66$ (c 1.55, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 7.31−7.20 (m, 8H), 7.16 (m, 2H), 5.88 (brs, 1H), 5.11 (dd, J = 13.1, 7.4 Hz, 1H), 5.04 (d, J = 12.3 Hz, 1H), 4.96 (d, J = 12.3 Hz, 1H), 4.02−3.89 (m, 4H), 3.27 (dd, J = 16.6, 7.4 Hz, 1H), 3.02 (dd, J = 23.3, 13.1 Hz, 1H), 2.97 (m, 1H), 2.90 (dd, J = 22.6, 13.1 Hz, 1H), 1.21 (t, J $= 7.0$ Hz, 3H), 1.15 (t, J = 6.8 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 200.0, 155.6, 141.3, 136.5, 128.6, 128.4, 128.0, 127.5, 126.3, 66.7, 62.9 (d, J = 5.3 Hz), 62.6 (d, J = 6.4 Hz), 51.3, 49.3, 43.5 (d, J = 123.8 Hz), 16.3 (d, $J = 4.1$ Hz), 16.2 (d, $J = 3.7$ Hz); HRMS-ESI (M + Na) m/z calcd for C₂₂H₂₈NO₆PNa 456.1552, found 456.1559.

General Procedure for the Synthesis of Enones 7. (R,E)-Ethyl-4-oxo-6-phenyl-hex-5-en-2-ylcarbamate $7a$. To a solution of 6a (0.5) g, 1.6 mmol, 1equiv) in THF (7 mL) was added in one time $Ba(OH)_2$ (0.346 g, 2.0 mmol, 1.25 equiv) at room temperature. After 30 min, a solution of benzaldehyde (0.172 mL, 1.7 mmol, 1.05 equiv) in THF/ H2O 40:1 (7 mL) was slowly added at room temperature. After 1 h, the reaction mixture was quenched with an aqueous solution of ammonium chloride and extracted three times with ethyl acetate. Then the organic layer was dried over Na_2SO_4 , concentrated under vacuum, and purified by flash chromatography (eluent, cyclohexane to cyclohexane/EtOAc 8:2) and gave compound 7a as a white solid $(0.40 \text{ g}, 95\%)$: mp 74 °C; $[\alpha]_D = +9.50$ (c 1.21, CHCl₃); ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 7.50 (d, J = 16.7 Hz, 1H), 7.47 (dd, J = 7.8, 3.0) Hz, 1H), 7.33−7.30 (m, 3H), 6.65 (d, J = 16.7 Hz, 1H), 5.14 (s, 1H), 4.14−4.06 (m, 1H), 4.02 (q, J = 6.9 Hz, 2H), 2.95 (dd, J = 15.9, 4.2 Hz, 1H), 2.71 (dd, $J = 15.9$, 6.5 Hz, 1H), 1.19 (d, $J = 6.8$ Hz, 3H), 1.14 (t, J = 6.9 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 198.8, 155.9, 143.4, 134.3, 130.6, 128.9, 128.4, 126.3, 60.6, 46.3, 44.1, 20.5, 14.6; HRMS-ESI $(M + Na)$ calcd for $C_{15}H_{19}NO_3Na$ 284.1263, found 284.1275.

(R,E)-Ethyl-6-(2-nitrophenyl)-4-oxohex-5-en-2-ylcarbamate 7b. $[\alpha]_{\text{D}}$ = +21.94 (c 1.015, CHCl₃); yellow solid; starting from 1.6 mmol of 6a, 0.43 g, yield = 89%; mp 90 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.07 (d, J = 8.1 Hz, 1H), 8.01 (d, J = 16.1 Hz, 1H), 7.71– 7.59 (m, 2H), 7.56 (t, J = 7.4 Hz, 1H), 6.58 (d, J = 16.1 Hz, 1H), 5.08 (brs, 1H), 4.22−4.08 (m, 3H), 3.04 (dd, J = 16.3, 4.7 Hz, 1H), 2.85 $(dd, J = 16.3, 6.3 Hz, 1H), 1.28 (d, J = 6.7 Hz, 3H), 1.22 (t, J = 7.0 Hz,$ 3H); ¹³C NMR (101 MHz, CDCl₃) δ 198.3, 155.8, 148.3, 138.7, 133.7, 131.1, 130.8, 130.5, 129.1, 125.1, 60.7, 46.0, 43.9, 20.6, 14.6; HRMS-ESI (M + Na) m/z calcd for $C_{15}H_{18}N_2O_5N_4$ 329.1113, found 329.1117.

(R,E)-Ethyl-6-(3-nitrophenyl)-4-oxohex-5-en-2-ylcarbamate 7c. $[\alpha]_{\text{D}}$ = +5.92 (c 0.995, CHCl₃); yellow solid; starting from 1.6 mmol of 6a, 0.41 g, yield = 86%; mp 97 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.43 (s, 1H), 8.27 (d, J = 8.1 Hz, 1H), 7.87 (d, J = 7.7 Hz, 1H), 7.65 (d, J = 16.2 Hz, 1H), 7.61 (d, J = 7.6 Hz, 1H), 6.85 (d, J = 16.2 Hz, 1H), 5.08 (brs, 1H), 4.27−4.04 (m, 3H), 3.07 (dd, J = 16.1, 3.3 Hz, 1H), 2.84 (dd, J = 16.1, 6.6 Hz, 1H), 1.31 (d, J = 6.7 Hz, 3H), 1.25 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 198.3, 155.6, 148.5, 140.2, 136.0, 133.9, 130.0, 128.6, 124.8, 122.6, 60.8, 47.0, 44.0, 20.5, 14.6; HRMS-ESI (M + Na) m/z calcd for $C_{15}H_{18}N_2O_5Na$ 329.1113, found 329.1125.

(R,E)-Ethyl-6-(4-nitrophenyl)-4-oxohex-5-en-2-ylcarbamate 7d. $\lbrack \alpha \rbrack_{\text{D}} = +16.42$ (c 0.52, CHCl₃); yellow solid; starting from 1.6 mmol of **6a**, 0.44 g, yield = 91%; mp 98 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.19 (d, J = 8.8 Hz, 2H), 7.62 (d, J = 8.5 Hz, 2H), 7.53 (d, J $= 16.2$ Hz, 1H), 6.75 (d, J = 16.2 Hz, 1H), 4.97 (s, 1H), 4.10 (m, 1H), 4.03 (q, J = 7.0 Hz, 2H), 2.99 (dd, J = 15.6, 3.5 Hz, 1H), 2.74 (dd, J = 15.6, 6.5 Hz, 1H), 1.21 (d, J = 6.8 Hz, 1H), 1.15 (t, J = 7.0 Hz, 1H); $13C$ NMR (101 MHz, CDCl₃) δ 198.0, 155.9, 140.5, 140.1, 128.5, 128.9, 124.8, 124.2, 60.8, 47.1, 44.0, 20.5, 14.6; HRMS-ESI (M + Na) m/z calcd for $C_{15}H_{18}N_2O_5Na$ 329.1113, found 329.1119.

(R,E)-Ethyl-6-(4-methoxyphenyl)-4-oxohex-5-en-2-ylcarbamate **7e.** $[\alpha]_D = +6.1$ (c 1.055, CHCl₃); yellow solid; starting from 1.6 mmol of 6a, 0.37 g, yield = 79%; mp 108 °C; ¹ H NMR (400 MHz, CDCl₃) δ 7.54 (d, J = 16.1 Hz, 1H), 7.50 (d, J = 7.9 Hz, 2H), 6.91 (d, J = 7.9 Hz, 1H), 6.60 (d, J = 16.1 Hz, 1H), 5.21 (brs, 1H), 4.22−4.07 (m, 3H), 3.83 (s, 3H), 2.99 (dd, J = 15.8, 4.4 Hz, 1H), 2.76 (dd, J = 15.8, 5.8 Hz, 1H), 1.25 (d, J = 6.8 Hz, 1H), 1.21 (t, J = 7.0 Hz, 1H); 13 C NMR (101 MHz, CDCl₃) δ 198.8, 161.9, 155.9, 143.2, 130.3, 127.1, 124.3, 114.6, 60.8, 55.5, 46.2, 44.4, 20.6, 14.7; HRMS-ESI (M + Na) m/z calcd for $C_{16}H_{21}NO_4$ Na 314.1368, found 314.1371.

(R,E)-Ethyl-6-(2-bromophenyl)-4-oxohex-5-en-2-ylcarbamate 7f. $[\alpha]_{\text{D}}$ = +18.2 (c 1.175, CHCl₃); yellow solid; starting from 1.6 mmol of 6a, 0.53 g, yield = 91%; mp 65 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.86 (d, J = 16.2 Hz, 1H), 7.56 (d, J = 7.8 Hz, 1H), 7.56 (d, J = 7.8 Hz, 1H), 7.27 (t, J = 7.8 Hz, 1H), 7.17 (td, J = 7.8, 1.5 Hz, 1H), 6.57 (d, J $= 16.2$ Hz, 1H), 5.08 (brs, 1H), 4.15−3.88 (m, 3H), 2.96 (dd, J = 16.4, 4.8 Hz, 1H), 2.79 (dd, J = 16.3, 6.2 Hz, 1H), 1.22 (d, J = 6.8 Hz, 3H), 1.16 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 198.8, 158.1, 145.0, 141.6, 133.5, 131.5, 129.0, 127.8, 60.7, 46.5, 44.0, 20.6, 14.6; HRMS-ESI (M + Na) m/z calcd for $C_{15}H_{18}BrNO_3Na$ 362.0368, found 362.0371.

(R,E)-Ethyl-6-(4-bromophenyl)-4-oxo-hex-5-en-2-ylcarbamate **7g.** $[\alpha]_D = +4.0$ (c 1.03, CHCl₃); yellow solid; starting from 1.6 mmol of 6a, 0.48 g, yield = 89%; mp 90 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.53 (d, J = 8.6 Hz, 2H), 7.50 (d, J = 16.2 Hz, 1H), 7.41 (d, J = 8.5 Hz, 1H), 6.70 (d, J = 16.2 Hz, 1H), 5.10 (s, 1H, NH), 4.14 (m, 1H), 4.09 $(q, J = 7.1 \text{ Hz}, 1H)$, 3.01 (dd, $J = 15.5$, 3.4 Hz, 1H), 2.77 (dd, $J = 15.5$, 6.6 Hz, 1H), 1.27 (d, $J = 6.8$ Hz, 1H), 1.23 (t, $J = 7.1$, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 198.4, 156.0, 141.9, 133.2, 132.2, 129.7, 126.7, 124.9, 60.7, 50.0, 46.5, 44.1, 20.5, 14.6; HRMS-ESI $(M + Na)$ m/z calcd for $C_{15}H_{18}BrNO_3Na$ 362.0368, found 362.0380.

(R,E)-Ethyl-6-(2-chloro-5-nitrophenyl)-4-oxo-hex-5-en-2-ylcarba*mate 7h.* $[\alpha]_D = +15.06$ (c 1.06, CHCl₃); yellow solid; starting from 1.6 mmol of 6a, 0.52 g, yield = 95%; mp 149 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.05 (d, J = 8.6 Hz, 1H), 7.97 (d, J = 16.0 Hz, 1H), 7.62 (s, 1H), 7.51 (d, $J = 8.6$, 1H), 6.57 (d, $J = 16.0$ Hz, 1H), 5.09 (brs, 1H), 4.23−4.03 (m, 3H), 3.05 (dd, J = 16.5, 5.2 Hz, 1H), 2.84 (dd, J = 16.5, 6.5 Hz, 1H), 1.28 (d, J = 6.9 Hz, 3H), 1.21 (t, J = 7.5 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 198.0, 158.9, 146.5, 140.4, 137.7, 132.9, 131.9, 130.5, 129.3, 126.7, 60.9, 46.5, 44.1, 20.7, 14.7; HRMS-ESI (M + Na) m/z calcd for $C_{15}H_{17}CIN_2O_5Na$ 363.0724, found 363.0717.

(R,E)-Ethyl-4-oxo-6-(pyridin-3-yl)hex-5-en-2-ylcarbamate 7i. $[\alpha]_{\text{D}}$ $= +10.25$ (c 0.865, CHCl₃); white solid; starting from 1.6 mmol of 6a, 0.36 g, yield = 87%; mp 90 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.71 $(d, J = 1.6 \text{ Hz}, 1H), 8.56 (d, J = 4.7 \text{ Hz}, 1H), 7.82 (dt, J = 7.9, 1.6, 1H),$ 7.51 (d, J = 16.3 Hz, 1H), 7.30 (dd, J = 7.9, 4.9 Hz, 1H), 6.72 (d, J = 16.3 Hz, 1H), 5.03 (brs, 1H), 4.20−3.70 (m, 3H), 2.98 (dd, J = 16.1, 4.0 Hz, 1H), 2.74 (dd, J = 16.1, 6.7 Hz, 1H), 1.21 (d, J = 6.9 Hz, 3H), 1.16 (t, J = 7.5 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 198.1, 155.9, 151.1, 149.9, 139.4, 134.5, 130.2, 128.0, 123.8, 60.7, 46.7, 44.1, 20.5, 14.6; HRMS-ESI (M + Na) m/z calcd for $C_{14}H_{18}N_2O_3N_4$ 285.1215, found 285.1210.

(R,2E,4E)-Ethyl-8-(ethoxycarbonylamino)-6-oxo-nona-2,4-dienoate 7j. $[\alpha]_{\text{D}} = +17.6$ (c 0.695, CHCl₃); viscous yellow oil; starting from 1.6 mmol of 6a, 0.38 g, yield = 85% ; ¹H NMR (400 MHz, CDCl₃) δ 7.24 (dd, J = 14.9, 11.4 Hz, 1H), 7.14 (dd, J = 15.1, 11.4 Hz, 1H), 6.35 (d, J = 15.0 Hz, 1H), 6.19 (d, J = 15.0 Hz, 1H), 4.97 (s, 1H), 4.17 (q, J = 7.1 Hz, 2H), 4.05−4.00 (m, 3H), 2.89 (d, J = 12.5 Hz, 1H), 2.67 (dd, J = 16.3, 6.5 Hz, 1H), 1.25 (t, J = 7.1 Hz, 3H), 1.18 (d, $J = 6.8$ Hz, 3H), 1.16 (t, $J = 7.1$ Hz, 3H); ¹³C NMR (101 MHz,

CDCl₃) δ 198.4, 165.7, 155.8, 141.1, 139.1, 135.3, 129.5, 60.9, 60.7, 46.7, 43.9, 20.4, 14.6, 14.2; HRMS-ESI (M + Na) m/z calcd for $C_{14}H_{21}NO_5Na$ 306.1317, found 306.1331.

(R)-Ethyl-6-(ethoxycarbonylamino)-4-oxohept-2-enoate 7k. Mixture of Z and E isomers $(Z/E = 60:40)$; colorless oil; starting from 1.6 mmol of 6a, 0.22 g, yield = 53%; ¹H NMR (400 MHz, CDCl₃) δ 6.96 $(d, J = 16.0 \text{ Hz}, 1H), 6.62 (d, J = 16.0 \text{ Hz}, 1H), 6.43 (d, J = 12.0 \text{ Hz},$ 1H), 5.97 (d, J = 12.0 Hz, 1H), 5.18 (s, 1H), 5.12 (s, 1H), 4.25−4.11 $(m, 6H)$, 2.91 (d, J = 15.0 Hz, 1H), 2.81 (dd, J = 16.9, 5.7 Hz, 1H), 2.78−2.67 (m, 1H), 1.27−1.13 (m, 12H); 13C NMR (101 MHz, CDCl3) (mixture of Z and E) δ 201.1, 197.3, 164.4, 164.2, 155.7, 155.0, 140.6, 138.3, 130.4, 124.0, 60.5, 60.3, 59.8, 59.7, 47.3, 46.0, 42.7, 42.4, 19.6, 19.4, 13.6, 13.1, 13.0; HRMS-ESI (M + Na) m/z calcd for $C_{12}H_{19}NO_5Na$ 280.1161, found 280.1163.

(R,E)-Ethyl-4-oxo-non-5-en-2-ylcarbamate 7l. $[\alpha]_{\text{D}} = +12.13$ (c 1.025, CHCl₃); colorless oil; starting from 1.6 mmol of $6a$, 0.34 g, yield = 95%; ¹H NMR (400 MHz, CDCl₃) δ 6.84 (dt, J = 15.9, 7.2 Hz, 1H), 6.07 (dd, J = 15.9, 1.5 Hz, 1H), 5.13 (brs, 1H), 4.11−4.02 (m, $3H$), 2.87 (dd, J = 16.1, 4.4 Hz, 1H), 2.65 (dd, J = 16.1, 6.4 Hz, 1H), 2.19 (td, $J = 7.2$, 1.5 Hz, 2H), 1.49 (qd, $J = 7.2$ Hz, 2H), 1.22 (t, $J = 7.2$ Hz, 3H), 1.21 (d, J = 6.8 Hz, 3H), 0.92 (t, J = 7.1 Hz, 3H); ¹³C NMR $(101 \text{ MHz}, \text{CDCl}_3)$ δ 199.1, 156.0, 148.5, 130.9, 60.7, 45.5, 44.2, 34.6, 21.4, 20.6, 14.7, 13.8; HRMS-ESI $(M + Na)$ m/z calcd for $C_{12}H_{21}NO_3Na$ 250.1419, found 250.1421.

 (R,E) -Ethyl-4-oxo-tetradec-5-en-2-ylcarbamate 7m. $\lceil \alpha \rceil_{\text{D}} =$ +10.71 (c 1.025, CHCl₃); yellow oil; starting from 1.6 mmol of $6a$, 0.42 g, yield = 88%; ¹H NMR (400 MHz, CDCl₃) δ 6.79 (dt, J = 15.8, 6.9 Hz, 1H), 6.01 (d, J = 15.8 Hz, 1H), 5.16 (s, 1H), 4.07−4.03 (m, 1H, 3H), 2.86 (dd, J = 16.0, 4.0 Hz, 1H), 2.63 (dd, J = 16.0, 6.4 Hz, 1H), 2.14 (q, J = 6.9 Hz, 2H), 1.45−1.32 (m, 2H), 1.30−1.18 (m, 15H), 1.15 (d, J = 7.1 Hz, 3H), 0.81 (t, J = 6.9 Hz, 3H); ¹³C NMR $(101 \text{ MHz}, \text{CDCl}_3)$ δ 199.1, 155.9, 148.7, 130.7, 60.7, 45.5, 44.1, 32.6, 31.9, 29.5, 29.4, 29.3, 29.2, 28.1, 22.7, 20.5, 14.7, 14.2; HRMS-ESI (M + Na) m/z calcd for C₁₇H₃₁NO₃Na 320.2202, found 320.2209.

(R,E)-Benzyl-4-oxo-6-phenyl-hex-5-en-2-ylcarbamate 7n. $[\alpha]_{\text{D}} =$ +2.56 (c 1.95, CHCl₃); white solid; starting from 1.3 mmol of 6b, 0.40 g, yield = 96%; mp 96 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.50 (d, J = 16.1 Hz, 1H), 7.48−7.46 (m, 2H), 7.33−7.19 (m, 8H), 6.64 (d, J = 16.1 Hz, 1H), 5.23 (s, 1H), 5.10−4.96 (m, 2H), 4.18−4.06 (m, 1H), 2.97 (d, J = 15.7 Hz, 1H), 2.73 (dd, J = 15.7, 5.6 Hz, 1H), 1.21 (d, J = 6.2 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 199.5, 155.6, 143.1, 134.0, 130.7, 129.0, 128.5, 128.4, 128.1, 126.3, 66.6, 46.1, 44.3, 20.5; HRMS-ESI (M + Na) m/z calcd for $C_{20}H_{21}NO_3Na$ 346.1419, found 346.1424.

(R,E)-Benzyl-4-oxo-pentadec-5-en-2-ylcarbamate 7o. $[\alpha]_{\text{D}} =$ -9.86 (c 0.975, CHCl₃); yellow oil; starting from 1.3 mmol of 6b, 0.38 g, yield = 78%; ¹H NMR (400 MHz, CDCl₃) δ 7.42–7.30 (m, 5H), 6.87 (dt, J = 16.0, 6.7 Hz, 1H), 6.09 (d, J = 16.0 Hz, 1H), 5.29 (brs, 1H), 5.10 (brs, 2H), 4.13 (m, 1H), 2.91 (dd, $J = 16.1$, 3.1 Hz, 1H), 2.69 (dd, $J = 16.1$, 6.1 Hz, 1H), 2.22 (dt, $J = 6.7, 7.1$ Hz, 2H), 1.52−1.41 (m, 2H), 1.36−1.27 (m, 12H), 1.25 (d, J = 6.8 Hz, 3H), 0.90 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 200.4, 155.6, 148.7, 139.5, 130.5, 128.5, 128.1, 128.0, 66.5, 45.2, 44.2, 32.5, 31.8, 29.5, 29.4, 29.3, 29.2, 28.04, 22.6, 20.4, 14.1; HRMS-ESI (M + H)⁺ m/ z calcd for $C_{23}H_{36}NO_3$ 374.2695, foun- 374.2706.

 (R, E) -Benzyl-4-oxo-heptadec-5-en-2-ylcarbamate 7p. $\lceil \alpha \rceil_{\text{D}} =$ -9.80 (c 1.015, CHCl₃); yellow oil; starting from 1.3 mmol of 6b, 0.45 g, yield = 91%; ¹H NMR (400 MHz, CDCl₃) δ 7.42–7.30 (m, 5H), 6.87 (dt, $J = 15.9$, 7.2 Hz, 1H), 6.06 (d, $J = 15.9$ Hz, 1H), 5.31 (brs, 1H), 5.08 (brs, 2H), 4.10 (m, 1H), 2.89 (d, J = 15.6 Hz, 1H), 2.66 (dd, J = 15.6, 5.3 Hz, 1H), 2.19 (q, J = 7.2 Hz, 2H), 1.47−1.40 (m, 6H), 1.33−1.21 (m, 15H), 0.88 (t, J = 6.9 Hz, 3H); 13C NMR $(101 \text{ MHz}, \text{CDCl}_3)$ δ 199.1, 155.7, 148.9, 136.7, 130.6, 128.6, 128.1, 66.6, 45.3, 44.3, 32.6, 32.0, 29.7, 29.6, 29.5, 29.4, 29.3, 28.1, 27.0, 22.8, 20.5, 14.2; HRMS-ESI (M + H) + m/z calcd for C₂₅H₄₀NO₃ 402.3008, found 402.3015.

(E)-Ethyl-6-oxo-8-phenyl-oct-7-en-4-ylcarbamate 7q and 7q′. R enantiomer: $[\alpha]_D$ = +21.87 (c 0.97, CHCl₃); S enantiomer: $[\alpha]_D$ = -21.32 (c 0.76 CHCl₃); white solid; starting from 1.5 mmol of 6c, 0.39 g, yield = 91%; mp 96 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.48

 $(d, J = 16.8 \text{ Hz}, 1H), 7.45-7.3 \text{ (m, 2H)}, 7.33-7.30 \text{ (m, 3H)}, 6.64 \text{ (d, J)}$ = 16.8 Hz, 1H), 5.06 (brs, 1H), 4.09−3.91 (m, 3H), 2.90 (dd, J = 17.2, 6.0 Hz, 1H), 2.78 (dd, J = 17.2, 5.5 Hz, 1H), 1.62−1.29 (m, 4H), 1.17 $(t, J = 6.9$ Hz, 3H), 0.85 $(t, J = 7.3$ Hz, 3H); ¹³C NMR (101 MHz, CDCl3) δ 199.0, 156.2, 143.2, 134.3, 130.6, 128.9, 128.4, 126.3, 60.6, 48.1, 44.8, 36.5, 19.5, 14.6, 13.8; HRMS-ESI $(M + Na)$ m/z calcd for $C_{17}H_{23}NO_3Na$ 312.1576, found 312.1585.

(R,E)-Ethyl-8-(4-nitrophenyl)-6-oxo-oct-7-en-4-ylcarbamate 7r. $[\alpha]_{\text{D}}$ = +13.8 (c 0.985, CHCl₃); yellow solid; starting from 1.5 mmol of 6c, 0.42 g, yield = 84%; mp 102 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.25 (d, J = 8.1 Hz, 2H), 7.70 (d, J = 8.1 Hz, 2H), 7.60 (d, J $= 16.2$ Hz, 1H), 6.83 (d, J = 16.2 Hz, 1H), 5.01 (s, 1H), 4.14–4.01 (m, 3H), 3.01 (d, J = 16.2 Hz, 1H), 2.85 (dd, J = 16.2, 5.7 Hz, 1H), 1.64− 1.49 (m, 2H), 1.47–1.30 (m, 2H), 1.22 (t, $J = 6.8$ Hz, 3H), 0.92 (t, $J =$ 7.2 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 198.5, 156.2, 140.6, 140.0, 129.5, 128.9, 124.2, 124.2, 60.8, 48.1, 45.8, 36.6, 19.5, 14.6, 13.8; HRMS-ESI (M + Na) m/z calcd for C₁₇H₂₂N₂O₅Na 357.1426, found 357.1420.

(R,E)-Ethyl-3-oxo-1-phenyl-tridec-1-en-5-ylcarbamate 7s. $[\alpha]_{\text{D}} =$ +17.26 (c 1.015, CHCl₃); white solid; starting from 1.2 mmol of 6d, 0.40 g, yield = 94%; mp 76 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.58 $(d, J = 16.1 \text{ Hz}, 1H), 7.57-7.52 \text{ (m, 2H)}, 7.44-7.36 \text{ (m, 3H)}, 6.74 \text{ (d,$ $J = 16.1$ Hz, 1H), 5.23 (d, J = 7.4 Hz, 1H), 4.21–3.94 (m, 3H), 3.01 $(d, J = 15.2 \text{ Hz}, 1\text{H})$, 2.84 $(dd, J = 15.2, 3.8 \text{ Hz}, 1\text{H})$, 1.69–1.49 (m, 2H), 1.48−1.11 (m, 15H), 0.85 (t, J = 6.8 Hz, 3H); 13C NMR (101 MHz, CDCl₃) δ 199.0, 156.2, 143.2, 134.3, 130.6, 129.6, 128.9, 128.4, 126.3, 60.6, 48.4, 44.9, 34.4, 31.8, 29.5, 29.2, 26.3, 22.6, 14.6, 14.1; HRMS-ESI (M + Na) m/z calcd for C₂₂H₃₃NO₃Na 382.2358, found 382.2364.

(R,E)-Ethyl-4-oxo-tetradec-2-en-6-ylcarbamate 7t. $[\alpha]_{\text{D}} = +17.44$ $(c \ 1.12, \ CHCl₃)$; white solid; starting from 1.2 mmol of 6d, 0.27 g, yield = 78%; mp 59 °C; ¹H NMR (400 MHz, CDCl₃) δ 6.87 (dt, J = 15.8, 6.8, 1H), 6.10 (d, $J = 15.8$, 1.6 Hz, 1H), 5.1 (brs, 1H), 4.07 (q, J $= 7.2$ Hz, 2H), 3.91 (m, 1H), 2.84 (dd, J = 15.8, 4.2 Hz, 1H), 2.67 (dd, J = 15.8, 5.5 Hz, 1H), 1.90 (d, 3H, J = 6.8 Hz), 1.55−1.46 (m, 2H), 1.33−1.20 (m, 15H), 0.86 (t, $J = 6.9$ Hz, 3H); ¹³C NMR (101 MHz, CDCl3) δ 199.2, 156.3, 143.6, 132.4, 60.7, 48.5, 44.0, 34.5, 31.9, 29.6, 29.5, 29.4, 26.4, 22.8, 18.5, 14.7, 14.2; HRMS-ESI (M + Na) m/z calcd for $C_{17}H_{31}NO_3Na$ 320.2202, found 320.2207.

(R,E)-Ethyl-6-oxo-hexadec-4-en-8-ylcarbamate 7u. $[\alpha]_{\text{D}} = +12.1$ (c 1.05, CHCl₃); colorless oil; starting from 1.2 mmol of 6d, 0.37 g, yield = 95%; ¹H NMR (400 MHz, CDCl₃) δ 6.83 (dt, J = 15.9, 7.2 Hz, 1H), 6.07 (dd, J = 15.9, 1.4 Hz, 1H), 5.11 (brs, 1H), 4.07 (qd, J = 6.9, 2H), 3.91 (m, 1H), 2.85 (dd, J = 16.3, 4.5 Hz, 1H), 2.69 (dd, J = 16.3, 5.7 Hz, 1H), 2.18 (dd, J = 7.2, 1.4 Hz, 1H), 1.53−1.44 (m, 2H), 1.49 $({\rm qd}, J = 7.2 \text{ Hz}, 2H), 1.29-1.19 \text{ (m, 15H)}, 0.92 \text{ (t, } J = 7.2 \text{ Hz}, 3H),$ 0.86 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 199.4, 156.3, 148.3, 130.9, 60.7, 46.5, 44.1, 34.6, 34.5, 31.9, 29.6, 29.5, 29.4, 26.4, 22.8, 21.4, 14.7, 14.2, 13.8; HRMS-ESI $(M + Na)$ m/z calcd for $C_{19}H_{35}NO_3Na$ 348.2515, found 348.2525.

(S,E)-Ethyl-3-oxo-1,5-diphenyl-pent-4-enylcarbamate $7v$. $[\alpha]_{D} =$ +6.6 (c 0.94, CHCl₃); yellow oil; starting from 1.3 mmol of 6e, 0.35 g, yield = 84%; ¹H NMR (400 MHz, CDCl₃) δ 7.43 (d, J = 16.2 Hz, 1H), 7.41−7.11 (m, 10H), 6.59 (d, J = 16.2 Hz, 1H), 5.74 (brs, 1H), 5.16 $(m, 1H)$, 4.01 $(q, J = 6.3 \text{ Hz}, 1H)$, 3.26 $(dd, J = 15.9 \text{ Hz}, 1H)$, 3.06 (dd, $J = 15.9$, 5.0 Hz, 1H), 1.13 (t, $J = 6.3$ Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 197.9, 155.9, 143.6, 134.2, 130.7, 128.9, 128.6, 128.4, 127.5, 126.3, 126.0, 61.0, 51.7, 46.1, 14.6; HRMS-ESI (M + Na) m/z calcd for $C_{20}H_{21}NO_3Na$ 346.1419, found 346.1414.

(S,E)-Ethyl-3-oxo-1-phenyl-hex-4-enylcarbamate 7w. $[\alpha]_{\text{D}} =$ -13.3 (c 0.715, CHCl₃); yellow oil; starting from 1.3 mmol of 6e, 0.27 g, yield = 80%; ¹H NMR (400 MHz, CDCl₃) δ 7.32–7.12 (m, 5H), 6.75 (dq, J = 15.9, 6.8 Hz, 1H), 5.96 (dd, J = 15.9,1.5 Hz, 1H), 5.68 (brs, 1H), 5.09 (m, 1H), 4.01 (q, J = 6.9 Hz, 2H), 3.12 (dd, J = 16.6, 5.8 Hz, 1H), 2.93 (dd, J = 16.6, 6.8 Hz, 1H), 1.79 (dd, J = 6.8, 1.5 Hz, 3H), 1.10 (t, J = 6.9 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 200.2, 158.2, 146.2, 143.7, 134.2, 130.8, 129.6, 128.5, 63.2, 53.9, 45.5, 20.6, 16.8; HRMS-ESI (M + Na) m/z calcd for C₁₅H₁₉NO₃Na 284.1263, found 284.1275.

(S,E)-Ethyl-3-oxo-1-phenyl-oct-4-enylcarbamate 7x. $[\alpha]_{\text{D}} =$ −9.11 (c 1.14, CHCl3); colorless oil; starting from 1.3 mmol of 6e, 0.36 g, yield = 95%; ¹H NMR (400 MHz, CDCl₃) δ 7.35–7.21 (m, 5H), 6.80 (dt, J = 15.9, 6.9 Hz, 1H), 6.05 (td, J = 15.9, 1.5, 1H), 5.75 $(s, 1H)$, 5.15 (dd, J = 5.8,3.1 Hz, 1H), 4.11 (q, J = 7.1 Hz, 2H), 3.23 $(dd, J = 16.1, 3.1 Hz, 1H), 3.04 (dd, J = 16.1, 5.8 Hz, 1H), 2.16 (gd, J)$ $= 6.9, 1.5$ Hz, 2H), 1.47 (sex, J = 6.8 Hz, 2H), 1.23 (t, J = 7.1 Hz, 3H), 0.92 (t, J = 6.8 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 198.3, 156.1, 148.8, 141.6, 130.6, 128.6, 127.4, 126.4, 61.0, 51.8, 45.3, 34.6, 21.3, 14.6, 13.7; HRMS-ESI $(M + Na)$ m/z calcd for $C_{17}H_{23}NO_3Na$ 312.1576, found 312.1576.

(S,E)-Ethyl-3-oxo-1-phenyl-tridec-4-enylcarbamate 7y. $[\alpha]_{\rm D}$ = -1.22 (c 0.995, CHCl₃); white solid; starting from 1.3 mmol of 6e, 0.41 g, yield = 89%; mp 60 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.23– 7.17 (m, 4H), 7.13 (m, 1H), 6.70 (dt, J = 15.9, 6.9 Hz, 1H), 5.93 (d, J = 15.9 Hz, 1H), 5.88 (brs, 1H), 5.07 (m, 1H), 3.98 (q, J = 7.1 Hz, 2H), 3.09 (d, J = 16.1 Hz, 1H), 2.89 (dd, J = 16.1, 5.7 Hz, 1H), 2.06 $(q, J = 6.9 \text{ Hz}, 2\text{H})$, 1.39–1.29 (m, 2H), 1.27–1.13 (m, 10H), 1.10 (t, $J = 7.1$ Hz, 3H), 0.79 (t, $J = 6.9$ Hz, 3H); ¹³C NMR (101 MHz, CDCl3) δ 198.2, 156.0, 148.9, 141.7, 130.3, 128.5, 127.3, 126.3, 60.8, 51.6, 45.3, 32.5, 31.8, 29.4, 29.3, 29.2, 29.1, 27.9, 22.6, 14.5, 14.1; HRMS-ESI (M + Na) m/z calcd for $C_{22}H_{33}NO_3Na$ 382.2358, found 382.2362.

(S,E)-tert-Butyl-3-oxo-1-phenyl-hex-4-enylcarbamate 7z. $[\alpha]_{\text{D}} =$ -10.68 (c 1.015, CHCl₃); white solid; starting from 1.2 mmol of 6f, 0.28 g, yield = 81%; mp 94 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.32– 7.18 (m, 5H), 6.84 (dq, J = 15.9, 6.8 Hz, 1H), 6.06 (dd, J = 15.9, 1.6 Hz, 1H), 5.55 (brs, 1H), 5.09 (m, 1H), 3.12 (d, $J = 16.6$ Hz, 1H), 2.98 (dd, J = 16.6, 5.6 Hz, 1H), 1.85 (dd, J = 6.8, 1.6 Hz, 3H), 1.40 (s, 9H); ¹³C NMR (101 MHz, CDCl₃) δ 198.2, 155.3, 144.0, 132.1, 128.7, 127.4, 126.4, 51.5, 45.6, 28.5, 18.5; HRMS-ESI (M + Na) m/z calcd for $C_{17}H_{23}NO_3Na$ 312.1576, found 312.1591.

(S,E)-Benzyl-3-oxo-1-phenyl-hex-4-enylcarbamate 1. $[\alpha]_{\text{D}} =$ -5.34 (c 1.05, CHCl₃); white solid; starting from 1.1 mmol of 6g, 0.3 g, yield = 96%; mp 60 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.42– 7.17 (m, 5H), 6.80 (dq, $J = 15.8$, 6.8 Hz, 1H), 6.05 (d, $J = 15.8$ Hz, 1H), 5.88 (brs, 1H), 5.18 (dd, $J = 6.2$, 5.6 Hz, 1H), 5.10 (d, $J = 12.3$ Hz, 1H), 5.05 (d, J = 12.3 Hz, 1H), 3.19 (dd, J = 16.2, 5.6 Hz, 1H), 3.00 (dd, $J = 16.2$, 6.2 Hz, 1H), 1.85 (d, $J = 6.8$ Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 197.8, 155.7, 144.0, 136.4, 136.2, 131.9, 128.7, 128.6, 128.5, 128.0, 126.4, 126.3, 66.8, 51.8, 45.1, 18.3; HRMS-ESI (M + Na) m/z calcd for $C_{21}H_{21}NO_3Na$ 346.1419, found 346.1426.

General Procedure for the Synthesis of Piperidines 8l−s/9l− s. (2S,6R)-Ethyl-4,4-dimethoxy-6-methyl-2-[(4-nitrophenyl)] piperidine-1-carboxylate 8m (trans isomer) and (2R,6R)-Ethyl-4,4 dimethoxy-6-methyl-2-[(4-nitrophenyl)]piperidine-1-carboxylate **9m** (cis isomer). In a one neck flash, to compound 7d (0.1 g, 0.32 mmol 1 equiv) were added successively, trimethyl orthoformate (1.75 mL, 1.60 mmol, 5 equiv) and p-toluenesulphonic acid (5.5 mg, 0.32 mmol, 1 equiv). The reaction was followed by TLC, and after 0.5 h ethyl acetate was added to the crude mixture, followed by a saturated solution of $NAHCO₃$ and extraction twice with ethyl acetate. The organic layer was dried and concentrated under vacuum before being purified by flash chromatography (eluent, cyclohexane to cyclohexane/ EtOAc 8:2) to yield a mixture of both isomer 8m and 9m (86 mg, 81% yield) in a ratio of 8m/9m 89/11 in favor of the trans isomer 8m.

8m. ¹H NMR (400 MHz, C_6D_6) δ 7.87 (d, J = 8.8 Hz, 2H), 6.84 (d, $J = 8.8$ Hz, 2H), 4.98 (t, $J = 5.3$ Hz, 1H), 4.26 (m, 1H), 4.09–3.92 (m, 2H), 2.87 (s, 3H), 2.58 (s, 3H), 2.03 (d, J = 5.3 Hz, 2H), 1.67 (dd, J = 3.5, 14.4 Hz, 1H),1.61 (dd, J = 5.5, 14.4 Hz, 1H), 1.37 (d, J = 6.7 Hz, 3H), 0.86 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, C_6D_6) δ 155.7, 150.8, 146.8, 123.3, 126.6, 98.2, 61.2, 53.7, 47.6, 47.2, 47.0, 37.6, 36.7, 20.6, 14.4; HRMS $(M + H)^+$ ion by direct probe calcd for $C_{17}H_{25}N_2O_6$ 353.1713, found 353.1699.

9m. ¹H NMR (400 MHz, C_6D_6) δ 7.84 (d, J = 8.8 Hz, 2H), 7.06 (d, J = 8.8 Hz, 2H), 5.19 (dd, J = 6.6, 6.9 Hz, 1H), 4.43 (m, 1H), 4.09− 3.92 (m, 2H), 2.98 (s, 3H), 2.79 (s, 3H), 2.01 (dd, $J = 6.9$, 14.1 Hz, 1H), 1.94 (dd, J = 6.6, 14.1 Hz, 1H), 1.79 (dd, J = 6.7, 14.2 Hz, 1H), 1.43 (dd, $J = 5.9$, 14.2 Hz, 1H), 1.16 (d, $J = 6.7$ Hz, 3H), 0.87 (t, $J =$

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7.1 Hz, 3H); ¹³C NMR (101 MHz, C_6D_6) δ 155.7, 151.9, 146.8, 123.8, 126.9, 98.6, 61.5, 54.0, 48.2, 47.2, 47.0, 36.9, 34.0, 23.0, 14.4.

(6R,2S)-Ethyl-4,4-dimethoxy-6-methyl-2-(pyridin-3-yl)piperidine-1-carboxylate 8k and (6R,2R)-Ethyl-4,4-dimethoxy-6-methyl-2- (pyridin-3-yl)piperidine-1-carboxylate 9k. Yellow oil, starting from 0.8 mmol of 7i, 0.20 g, yield = 81% in a ratio of $8k/9k$ 85/15 in favor of the trans isomer 8k.

8k. ¹H NMR (400 MHz, CDCl₃) δ 8.40 (m, 1H), 8.38 (d, J = 4.5 Hz, 1H), 7.44 (d, J = 7.3 Hz, 1H), 7.16 (dd, J = 7.3, 4.5 Hz, 1H), 5.09 (t, J = 5.3 Hz, 1H), 4.26 (m, 1H), 4.06−3.94 (m, 2H), 3.11 (s, 3H), 2.85 (s, 3H), 2.35 (dd, J = 14.4, 5.3 Hz, 1H), 2.29 (dd, J = 14.4, 5.3 Hz, 1H), 1.88 (dd, J = 14.4, 5.3 Hz, 1H), 1.82 (dd, J = 14.4, 5.5 Hz, 1H), 1.32 (d, $J = 6.8$ Hz, 3H), 1.04 (t, $J = 7.1$ Hz, 3H); ¹³C NMR (101) MHz, CDCl₃) δ 156.1, 147.7, 147.5, 137.9, 133.5, 122.9, 98.2, 61.3, 51.7, 47.8, 47.6, 46.8, 37.3, 36.6, 20.6, 14.4.

9k. ¹H NMR (400 MHz, CDCl₃) δ 8.53 (m, 1H), 8.37 (d, J = 4.5 Hz, 1H), 7.55 (d, J = 7.3 Hz, 1H), 7.18 (dd, J = 7.3, 4.5 Hz, 1H), 5.16 (dd, J = 7.1, 6.4 Hz, 1H), 4.35 (m, 1H), 4.06–3.94 (m, 2H), 3.16 (s, 3H), 3.05 (s, 3H), 2.27 (m, 1H), 2.15 (dd, J = 14.6, 6.4 Hz, 1H), 2.00 $(dd, J = 14.4, 6.9 \text{ Hz}, 1 \text{H}$, 1.66 (dd, J = 14.4, 5.5 Hz, 1H), 1.22 (d, J = 6.9 Hz, 3H), 1.04 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 155.9, 147.4, 147.9, 138.1, 133.8, 123.1, 98.4, 61.5, 52.3, 47.9, 47.6, 47.3, 37.4, 36.7, 22.9, 14.4; HRMS-ESI (M + Na) m/z calcd for $C_{16}H_{24}N_2O_4$ Na 331.1634, found 331.1634.

(2S,6R)-ethyl-4,4-dimethoxy-6-propyl-2-[(4-nitrophenyl)] piperidine-1-carboxylate 8l and 9l. Yellow oil, starting from 0.6 mmol of 7r, 0.16 g, yield = 71% in a ratio of $81/9183/17$ in favor of the trans isomer 8l.

8l. ¹H NMR (400 MHz, C_6D_6) δ 7.83 (d, J = 8.7 Hz, 2H), 6.84 (d, J $= 8.7$ Hz, 2H), 4.87 (t, J = 5.4 Hz, 1H), 3.99 (m, 1H), 4.09–3.92 (q, J $= 7.1$ Hz, 2H), 2.82 (s, 3H), 2.56 (s, 3H), 1.95 (d, J = 5.4 Hz, 2H), 1.78 (dd, $J = 14.6$, 4.0 Hz, 1H), 1.58 (m, 1H), 1.51 (dd, $J = 14.6$, 5.3 Hz, 1H), 1.37–1.25 (m, 3H), 0.97 (t, J = 7.5 Hz, 3H), 0.81 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, C_6D_6) δ 155.9, 150.5, 146.8, 126.7, 123.2, 98.7, 61.2, 53.6, 52.0, 47.4, 47.0, 37.0, 36.4, 34.1, 20.6, 14.5, 14.1; HRMS-ESI (M + Na) m/z calcd for C₁₉H₂₈N₂O₆Na 403.1845, found 403.1831

(2S,6R)-Ethyl-4,4-dimethoxy-6-methyl-2-[(4-nitrophenyl)] piperidine-1-carboxylate 8m and (2R,6R)-Ethyl-4,4-dimethoxy-6 methyl-2-[(4-nitrophenyl)]piperidine-1-carboxylate 9m. Yellow oil, starting from 1.0 mmol of 7d, 0.28g, yield =80% in a ratio of 8m/9m 89/11 in favor of the trans isomer 8m.

8m. ¹H NMR (400 MHz, C_6D_6) δ 7.87 (d, J = 8.8 Hz, 2H), 6.84 (d, J = 8.8 Hz, 2H), 4.98 (t, J = 5.3 Hz, 1H), 4.26 (m, 1H), 4.09−3.92 (m, 2H), 2.87 (s, 3H), 2.58 (s, 3H), 2.03 (d, $J = 5.3$ Hz, 2H), 1.67 (dd, $J =$ 14.4, 3.5 Hz, 1H), 1.61 (dd, J = 14.4, 5.5 Hz, 1H), 1.37 (d, J = 6.7 Hz, 3H), 0.86 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, C_6D_6) δ 155.7, 150.8, 146.8, 123.3, 126.6, 98.2, 61.2, 53.7, 47.6, 47.2, 47.0, 37.6, 36.7, 20.6, 14.4.

9m. ¹H NMR (400 MHz, C_6D_6) δ 7.84 (d, J = 8.8 Hz, 2H), 7.06 (d, J = 8.8 Hz, 2H), 5.19 (dd, J = 6.9, 6.6 Hz, 1H), 4.43 (m, 1H), 4.09− 3.92 (m, 2H), 2.98 (s, 3H), 2.79 (s, 3H), 2.01 (dd, $J = 14.1$, 6.9 Hz, 1H), 1.94 (dd, J = 14.1, 6.6 Hz, 1H), 1.79 (dd, J = 14.2, 6.7 Hz, 1H), 1.43 (dd, $J = 14.2, 5.9$ Hz, 1H), 1.16 (d, $J = 6.7$ Hz, 3H), 0.87 (t, $J =$ 7.1 Hz, 3H); ¹³C NMR (101 MHz, C_6D_6) δ 155.7, 151.9, 146.8, 123.8, 126.9, 98.6, 61.5, 54.0, 48.2, 47.2, 47.0, 36.9, 34.0, 23.0, 14.4; HRMS $(M + H)^+$ ion by direct probe calcd for $C_{17}H_{25}N_2O_6$ 353.1713, found 353.1699.

(2S,6R)-Ethyl-4,4-dimethoxy-6-methyl-2-[(3-nitrophenyl)] piperidine-1-carboxylate 8n and (2R,6R)-Ethyl-4,4-dimethoxy-6 methyl-2-[(3-nitrophenyl)]piperidine-1-carboxylate 9n. Yellow oil, starting from 1.0 mmol of 7c, 0.25 g, yield = 71% in a ratio of $8n/9n$ 40/60 in favor of the cis isomer 9n.

8n. ¹H NMR (400 MHz, C₆D₆) δ 8.21 (m, 1H), 7.87 (d, J = 8.0 Hz, 1H), 7.24 (d, J = 7.8 Hz, 1H), 6.91 (dd, J = 8.0, 7.8 Hz, 1H), 4.98 (dd, J = 5.3, 4.5 Hz, 1H), 4.26 (m, 1H), 4.18−4.10 (m, 2H), 2.96 (s, 3H), 2.71 (s, 3H), 2.18 (dd, J = 14.4, 4.5 Hz, 1H), 2.13 (dd, J = 14.4, 5.3 Hz, 1H), 1.67 (dd, J = 14.4, 4.0 Hz, 1H), 1.73 (dd, J = 14.4, 4.8 Hz, 1H), 1.47 (d, J = 6.8 Hz, 3H), 1.02 (t, J = 7.1 Hz, 3H); ¹³C NMR (101)

MHz, C_6D_6) δ 155.8, 148.7, 145.6, 131.6, 128.8, 121.3, 121.1, 98.2, 61.2, 53.5, 47.5, 47.2, 47.0, 37.4, 36.8, 20.6, 14.5.

9n. ¹H NMR (400 MHz, C_6D_6) δ 8.40 (m, 1H), 7.87 (d, J = 8.0 Hz, 1H), 7.45 (d, J = 7.8 Hz, 1H), 6.91 (dd, J = 8.0, 7.8 Hz, 1H), 5.40 (dd, J = 6.8, 6.6 Hz, 1H), 4.55 (qd, J = 6.8, 5.8 Hz, 1H), 4.13−4.04 (m, 2H), 3.02 (s, 3H), 2.90 (s, 3H), 2.19 (dd, $J = 14.6$, 6.8 Hz, 1H), 2.06 $(ddd, J = 14.6, 6.6, 0.7 Hz, 1H), 1.90 (ddd, J = 14.1, 6.8, 0.7 Hz, 1H),$ 1.55 (dd, J = 14.4, 5.8 Hz, 1H), 1.29 (d, J = 6.8 Hz, 3H), 0.99 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, C₆D₆) δ 156.3, 148.7, 146.8, 132.2, 129.0, 121.5, 121.4, 98.6, 61.5, 54.0, 47.6, 47.6, 47.2, 37.4, 36.7, 23.0, 14.5; HRMS $(M + H)^+$ ion by direct probe calcd for $C_{17}H_{25}N_2O_6$ 353.1713, found 353.1710.

(2S,6R)-Ethyl-4,4-dimethoxy-6-methyl-2-[(2-nitrophenyl)] piperidine-1-carboxylate 8o and (2R,6R)-Ethyl-4,4-dimethoxy-6 methyl-2-[(2-nitrophenyl)]piperidine-1-carboxylate 90. Yellow oil, starting from 1.0 mmol of 7b, 0.23 g, yield = 67 % in a ratio of $8o/9o$ 60/40 in favor of the trans isomer 8o.

80. ¹H NMR (400 MHz, C₆D₆) δ 7.51 (dd, J = 8.0, 1.2 Hz, 1H), 7.50 (d, J = 7.8 Hz, 1H), 7.07 (td, J = 8.0, 1.0 Hz, 1H), 6.77 (ddd, J = 8.0, 7.8, 1.2 Hz, 1H), 5.59 (dd, J = 12.4, 5.3 Hz, 1H), 4.72 (qtd, J = 6.8, 4.5 Hz, 1H), 3.91−3.71 (m, 2H), 3.2 (s, 3H), 3.01 (s, 3H), 2.18 (ddd, $J = 14.4, 5.3, 1.2$ Hz, 1H), 2.80 (ddd, $J = 14.4, 6.8, 1.2$ Hz, 1H), 1.95 $(dd, J = 14.4, 12.4 Hz, 1H), 1.71 (dd, J = 14.4, 4.5 Hz, 1H), 1.52 (d, J =$ 6.8 Hz, 3H), 0.74 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, C_6D_6) δ 156.1, 149.6, 141.5, 132.6, 127.6, 126.9, 124.5, 98.4, 61.2, 52.2, 47.9, 47.7, 47.0, 37.9, 36.9, 24.0, 13.9.

90. ¹H NMR (400 MHz, C_6D_6) δ 7.55 (dd, J₇= 8.0, 1.2 Hz, 1H), 7.32 (d, J = 7.8 Hz, 1H), 7.03 (td, J = 8.0, 1.0 Hz, 1H), 6.76 (ddd, J = 8.0, 7.8, 1.2 Hz, 1H), 5.94 (dd, J = 6.6, 5.3 Hz, 1H), 4.55 (m, 1H), 4.02−3.84 (m, 2H), 3.00 (s, 3H), 2.79 (s, 3H), 2.40 (dd, J = 14.4, 5.3 Hz, 1H), 2.21 (dd, $J = 14.4$, 6.6 Hz, 1H), 1.86 (dd, $J = 14.4$, 5.5 Hz, 1H), 1.80 (dd, $J = 14.4$, 3.3 Hz, 1H), 1.47 (d, $J = 6.8$ Hz, 3H), 0.89 (t, $J = 7.1$ Hz, 3H); ¹³C NMR (101 MHz, C₆D₆) δ 156.0, 149.2, 139.3, 132.0, 127.8, 127.0, 123.8, 98.1, 61.1, 50.6, 48.5, 47.7, 47.4, 38.8, 37.4, 20.2, 14.3; HRMS $(M + H)^+$ ion by direct probe calcd for $C_{17}H_{25}N_2O_6$ 353.1713, found 353.1711.

(2S,6R)-ethyl-2-(4-bromophenyl)-4,4-dimethoxy-6-methylpiperidine-1-carboxylate 8p and (2R,6R)-ethyl-2-(4-bromophenyl)-4,4 dimethoxy-6-methylpiperidine-1-carboxylate 9p. Yellow oil, starting from 0.9 mmol of 7g, 0.25 g, yield = 73% in a ratio of $8p/9p$ 33/67 in favor of the *cis* isomer 9p.

8p. ¹H NMR (400 MHz, C_6D_6) δ 7.13 (d, J = 8.3 Hz, 2H), 6.68 (d, $J = 8.3$ Hz, 2H), 4.91 (dd, $J = 5.3$, 4.5 Hz, 1H), 4.15 (m, 1H), 3.91 (qd, $J = 7.1$ Hz, 2H), 2.75 (s, 3H), 2.49 (s, 3H), 2.10 (dd, $J = 14.4$, 4.5 Hz, 1H), 2.03 (dd, J = 14.4, 5.3 Hz, 1H), 1.82 (dd, J = 14.4, 5.3 Hz, 1H), 1.78 (dd, $J = 14.4$, 3.8 Hz, 1H), 1.40 (d, $J = 6.8$ Hz, 3H), 0.89 (t, $J =$ 7.1 Hz, 3H); ¹³C NMR (101 MHz, C_6D_6) δ 155.9, 142.5, 131.3, 128.0, 120.1, 98.5, 61.0, 53.5, 47.7, 47.5, 47.4, 37.5, 36.7, 20.6, 14.6.

9p. ¹H NMR (400 MHz, C_6D_6) δ 7.12 (d, J = 8.6 Hz, 2H), 6.92 (d, $J = 8.6$ Hz, 2H), 5.13 (t, $J = 6.8$ Hz, 1H), 4.32 (sex, $J = 6.6$ Hz, 1H), 3.88 (qd, J = 7.1 Hz, 2H), 2.79 (s, 3H), 2.69 (s, 3H), 2.09 (dd, J = 14.6, 6.8 Hz, 1H), 2.01 (ddd, J = 14.6, 6.8, 0.7 Hz, 1H), 1.82 (ddd, J = 14.1, 6.6, 0.7 Hz, 1H), 1.49 (dd, $J = 14.1$, 6.6 Hz, 1H), 1.18 (d, $J = 6.6$ Hz, 3H), 0.87 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, C_6D_6) δ 156.5, 143.8, 131.5, 128.3, 120.5, 98.8, 61.3, 53.8, 47.2, 47.0, 37.6, 36.9, 23.0, 14.5; HRMS-ESI (M + Na) m/z calcd for C₁₇H₂₄NO₄BrNa 408.0786, found 408.0794

(2S,6R)-ethyl-2-(4-bromophenyl)-4,4-dimethoxy-6-methylpiperidine-1-carboxylate 8q and (2R,6R)-ethyl-2-(4-bromophenyl)-4,4 dimethoxy-6-methylpiperidine-1-carboxylate 9q. Yellow oil, starting from 0.9 mmol of 7f, 0.26 g, yield = 75% in a ratio of $8q/9q$ 29/71 in favor of the cis isomer 9q.

8q. ¹H NMR (400 MHz, C_6D_6) δ 7.51 (dd, J = 7.7, 1.2 Hz, 1H), 7.47 (dd, J = 7.7, 0.7 Hz, 1H), 7.09 (t, J = 7.7 Hz, 1H), 6.79 (t, J = 7.7 Hz, 1H), 5.58 (dd, J = 12.4, 5.1 Hz, 1H), 4.85 (qtd, J = 7.0, 3.7 Hz, 1H), 3.91−4.08 (m, 2H), 3.15 (s, 3H), 2.97 (s, 3H), 2.57 (dd, J = 14.3, 5.1 Hz, 1H), 2.11 (dd, J = 14.1, 7.5 Hz, 1H), 1.93 (dd, J = 14.3, 12.4 Hz, 1H), 1.72 (dd, $J = 14.1$, 3.7 Hz, 1H), 1.40 (d, $J = 7.0$ Hz, 3H), 0.83 $(t, J = 7.1 \text{ Hz}, 3\text{H})$. ¹³C NMR (101 MHz, C₆D₆) δ 156.8, 133.0, 129.7,

128.3, 127.9, 126.8, 98.6, 61.1, 56.0, 47.5, 47.4, 47.3, 39.0, 37.4, 24.1, 14.1.

9q. ¹H NMR (400 MHz, C_6D_6) δ 7.77 (d, J = 8.2 Hz, 1H), 7.46 $(dd, J = 7.7, 0.7 Hz, 1H), 7.46 (t, J = 7.7 Hz, 1H), 6.76 (m, 1H), 5.56$ $(dd, J = 5.8, 5.4 Hz, 1H), 4.69 (m, 1H), 3.90 (m, 2H), 3.03 (s, 3H),$ 2.80 (s, 3H), 2.62 (dd, J = 14.5, 5.4 Hz, 1H), 2.33 (dd, J = 14.5, 5.8 Hz, 1H), 2.05 (dd, J = 14.6, 6.2 Hz, 1H), 1.89 (m, 1H), 1.57 (d, J = 6.9 Hz, 3H), 0.91 (t, J = 6.9 Hz, 3H); ¹³C NMR (101 MHz, C_6D_6) δ 156.3, 133.1, 128.6, 128.3, 127.9, 127.2, 98.5, 60.9, 54.6, 48.2, 48.0, 47.0, 37.2, 36.2, 20.7, 14.4; HRMS-ESI (M + Na) m/z calcd for $C_{17}H_{24}NO_4BrNa$ 408.0786, found 408.0790

(2S,6R)-ethyl-4,4-dimethoxy-6-methyl-2-[(3-nitro-5 chlorophenyl)]piperidine-1-carboxylate 8r and (2S,6R)-ethyl-4,4 dimethoxy-6-methyl-2-[(3-nitro-5-chlorophenyl)]piperidine-1-carboxylate 9r. Yelow oil, starting from 1.2 mmol of 7h, 0.35 g; yield = 76% in a ratio of 8r/9r 56/43 in favor of the trans isomer 8r.

8r. ¹H NMR (400 MHz, C_6D_6) δ 7.60 (d, J = 8.6 Hz, 1H), 7.56 (d, J $= 2.3$ Hz, 1H), 6.60 (dd, J = 8.6, 2.3 Hz, 1H), 5.47 (dd, J = 12.1, 5.3 Hz, 1H), 4.49 (qtd, J = 6.8, 4.8 Hz, 1H), 3.77−3.58 (m, 2H), 3.03 (s, 3H), 2.87 (s, 3H), 2.59 (ddd, J = 14.4, 5.3, 1.5 Hz, 1H), 1.89 (ddd, J = 14.4, 6.8, 1.5 Hz, 1H), 1.75 (dd, J = 14.4, 12.1 Hz, 1H), 1.48 (dd, J = 14.4, 4.8 Hz, 1H), 1.32 (d, J = 6.8 Hz, 3H), 0.63 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, C₆D₆) δ 156.0, 147.7, 143.8, 139.0, 127.9, 127.2, 125.6, 98.2, 61.4, 52.5, 47.9, 48.5, 47.5, 38.6, 37.5, 19.2, 13.9.

9r. ¹H NMR (400 MHz, C_6D_6) δ 7.44 (d, J = 2.3 Hz, 1H), 6.6 (d, J $= 8.6$ Hz, 1H), 6.59 (dd, J = 8.6, 2.3 Hz, 1H), 5.70 (dd, J = 6.8, 5.3 Hz, 1H), 4.30 (m, 1H), 3.90−3.75 (m, 2H), 2.86 (s, 3H), 2.68 (s, 3H), 2.29 (dd, $J = 14.4$, 5.3 Hz, 1H), 1.96 (dd, $J = 14.4$, 6.8 Hz, 1H), 1.68 $(dd, J = 14.4, 5.5 Hz, 1H), 1.64 (dd, J = 14.4, 3.3 Hz, 1H), 1.28 (d, J =$ 6.8 Hz, 3H), 0.78 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, C_6D_6) δ 156.0, 147.7, 141.7, 138.6, 129.7, 127.2, 126.2, 98.0, 61.3, 50.5, 47.7, 47.5, 47.1, 37.3, 36.9, 24.0, 14.3; HRMS $(M + H)^+$ ion by direct probe calcd for $C_{17}H_{24}CIN_2O_6$ 387.1323, found 387.1325.

(2S,6R)-ethyl-2-((E)-3′-ethoxy-3′-oxoprop-1′-enyl)-4,4-dimethoxy-6-methylpiperidine-1-carboxylate 8s. Yellow oil, starting from 0.7 mmol of 7j, 0.21 g, yield = 93% in a ratio of 8s/9s: 90/10 in favor of the trans isomer 8s.

8s. ¹H NMR (400 MHz, CDCl₃) δ 6.89 (dd, J = 15.7, 5.3 Hz, 1H), 5.71 (dd, $J = 15.7, 2.0$ Hz, 1H), 4.60 (tdd, $J = 5.3, 4.2, 2.0$ Hz, 1H), 4.14−4.02 (m, 5H), 3.12 (s, 3H), 3.06 (s, 3H), 2.13 (dd, J = 14.2, 5.3 Hz, 1H), 2.06 (dd, J = 14.2, 4.2 Hz, 1H), 1.92 (dd, J = 14.4, 4.5 Hz, 1H), 1.83 (dd, J = 14.4, 3.3 Hz, 1H), 1.24−1.17 (m, 9H); ¹³C NMR $(101 \text{ MHz}, \text{ C}_6\text{D}_6) \delta$ 166.1, 155.4, 149.4, 141.2, 98.2, 61.2, 60.1, 51.7, 47.3, 47.2, 47.0, 37.1, 36.8, 20.7, 14.7, 14.2; HRMS-ESI $(M + Na)$ m/z calcd for $C_{16}H_{27}NO_6N$ a 352.1736, found 352.1729.

General Procedure for the Synthesis of Piperidines 10/11. (7R,9S)-Ethyl-7-methyl-9-phenyl-1,4-dithia-8-azaspiro[4.5]decane-8-carboxylate 10c and (7S,9S)-Ethyl-7-methyl-9-phenyl-1,4-dithia-8-azaspiro[4.5]decane-8-carboxylate 11c. To a solution of compound 7w (100 mg, 0.38 mmol, 1 equiv) were added successively trimethyl orthoformate (0.21 mL, 1.90 mmol, 5 equiv) and ptoluenesulphonic acid PTSA/ H_2O (1.5 mg, 0.08 mmol, 0.2 equiv). The reaction was followed by TLC, and after 0.5 h ethyl acetate was added to the crude mixture, quenched with a saturated solution of $NaHCO₃$ and extracted twice with ethyl acetate. Then the organic layer was dried and concentrated under vacuum. After ¹H NMR spectroscopy for identifying the product and measuring the de, the crude mixture 8c/9c (0.117 g, 0.38 mmol, ratio 8c/9c 44/56) was engaged in reaction without any purification. To a solution of 8c/9c in dry CH_2Cl_2 (5 mL) were added successively 1,2-ethandithiol (161 mL, 1.9 mmol, 5 equiv) and BF_3 ·Et₂O (243 mL, 1.9 mmol, 5 equiv) dropwise at 0 °C. After 2 h at room temperature, the solution was quenched dropwise with a solution of NaOH at 0 °C and extracted three times with ethyl acetate. Then the organic layer was dried over Na2SO4 and concentrated under vacuum before being purified by flash chromatography (eluent, cyclohexane to cyclohexane/EtOAc 9:1) to give a mixture of both isomer 10c and 11c (0.088 g, 68%), ratio $10c/$ 11c 44/56). HRMS-ESI (M + Na) m/z calcd for C₁₇H₂₃NO₂S₂Na 360.1068, found 360.1069.

10c. ¹ H NMR (400 MHz, CDCl3) δ 7.36−7.12 (m, 5H), 5.26 (dd, J $= 4.8, 5.0$ Hz, 1H), 4.41 (m, 1H), 3.97 (q, J = 7.1 Hz, 2H), 3.14–3.01 $(m, 4H)$, 2.80 (dd, J = 5.0, 14.9 Hz, 1H), 2.74 (dd, J = 4.8, 14.9 Hz, 1H), 2.51 (dd, J = 5.0, 14.9 Hz, 1H), 2.22 (dd, J = 4.3, 14.9 Hz, 1H), 1.43 (d, $J = 7.1$ Hz, 3H), 1.03 (t, $J = 7.1$ Hz, 3H); ¹³C NMR (101) MHz, CDCl₃) δ 156.2, 141.2, 128.2, 126.7, 125.9, 61.1, 61.0, 55.2, 48.6, 46.3, 46.0, 39.3, 39.2, 20.5, 14.5

11c. ¹H NMR (400 MHz, CDCl₃) δ 7.36–7.12 (m, 5H), 5.26 (dd, J $= 7.8, 8.3, Hz, 1H$, 4.41 (tdd, J = 7.0, 7.5, 6.8 Hz, 1H), 4.03 (q, J = 7.1) Hz, 2H), 3.26−3.21 (m, 4H), 2.63 (ddd, J = 7.8, 14.6, 1.8 Hz, 1H), 2.57 (dd, J = 8.3, 14.6 Hz, 1H), 2.44 (ddd, J = 7.0, 14.4, 1.8 Hz, 1H), 2.03 (dd, $J = 7.5$, 14.4 Hz, 1H), 1.22 (d, $J = 6.8$ Hz, 3H), 1.06 (t, $J =$ 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 156.7, 143.7, 128.3, 126.6, 125.9, 62.4, 61.4, 55.7, 48.2, 46.2, 45.5, 39.2, 39.1, 23.5, 14.5.

(7R,9S)-7-Methyl-9-phenyl-1,4-dithia-8-azaspiro[4.5]decane 10a and (7R,9R)-7-Methyl-9-phenyl-1,4-dithia-8-azaspiro[4.5]decane **11a.** Yellow oil, starting from 0.8 mmol of 7a, 0.15 g, yield = 68% in a ratio of 10a/11a 86/14 in favor of the trans isomer 10a.

10a. ¹H NMR (400 MHz, CDCl₃) δ 7.38–7.25 (m, 5H), 4.18 (dd, J = 8.8, 3.4 Hz, 1H), 3.47 (m, 1H), 3.35−3.10 (m, 4H), 2.38 (ddd, J = 13.7, 5.2 Hz, 1H), 2.28 (ddd, J = 13.4, 3.4, 1.6 Hz, 1H), 2.18 (dd, J = 13.3, 8.8 Hz, 1H), 1.96 (ddd, J = 13.7, 4.0, 1.6 Hz, 1H), 1.52 (s, 1H), 1.30 (t, J = 7.0 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 143.5, 128.4, 127.0, 126.7, 65.2, 53.8, 49.9, 48.1, 45.8, 39.6, 37.7, 20.8; HRMS (M + H ⁺ ion by direct probe calcd for $C_{14}H_{19}NS_2$ 266.1037, found 266.1042

11a. ¹H NMR (400 MHz, CDCl₃) δ 7.38–7.25 (m, 5H), 3.85 (dd, $J = 11.2, 2.3$ Hz, 1H), 3.30–3.20 (m, 4H), 2.95 (m, 1H), 2.15 (dt, $J =$ 13.1, 2.3 Hz, 1H), 2.05 (dt, J = 12.9, 2.3 Hz, 1H), 1.95 (dd, J = 13.3,11.2 Hz, 1H), 1.75 (dd, J = 12.9, 11.0 Hz, 1H), 1.08 (d, J = 7.0 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 142.8, 127.4, 126.3, 125.8, 65.9, 59.9, 51.0, 49.4, 48.7, 38.2, 36.7, 21.3; HRMS (M + H)⁺ ion by direct probe calcd for $C_{14}H_{19}NS_2$ 266.1037, found 266.1040.

(7S,9R)-Ethyl-7-phenyl-9-propyl-1,4-dithia-8-azaspiro[4.5] decane-8-carboxylate 10f and (7R,9R)-Ethyl-7-phenyl-9-propyl-1,4 dithia-8-azaspiro[4.5]decane-8-carboxylate 11f. Yellow oil, starting from 1.0 mmol of 7q, 0.22 g, yield = 60% (entry 6, Table 1) in a ratio of $10f/11f \geq 95/5$ in favor of the *trans* isomer 10f.

10f. ¹H NMR (400 MHz, CDCl₃) δ 7.26−7.12 (m, 5H), 5.20 (dd, J $= 9.3, 7.3$ Hz, 1H[\),](#page-2-0) 4.29 (m, 1H), 4.02 (q, J = 7.1 Hz, 2H), 3.28–3.20 $(m, 4H)$, 2.56 (ddd, J = 14.4, 7.31, 1.8 Hz, 1H), 2.53 (dd, J = 14.4, 9.3 Hz, 1H), 2.50 (ddd, $J = 14.1$, 8.4, 1.8 Hz, 1H), 2.11 (dd, $J = 14.1$, 5.8 Hz, 1H), 1.67 (m, 1H), 1.42−1.21 (m, 3 H), 1.04 (t, J = 6.9 Hz, 3H), 0.82 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 157.0, 143.8, 128.2, 126.6, 125.9, 62.4, 61.4, 55.8, 52.1, 45.9, 44.2, 39.2, 39.1, 19.7, 14.4, 13.8.

11f. ¹ H NMR (400 MHz, CDCl3) δ 7.26−7.12 (m, 5H), 5.09 (dd, J = 5.5, 5.2 Hz, 1H), 4.04−3.95 (m, 3H), 3.17−3.03 (m, 4H), 2.75 (dd, $J = 14.6, 5.2$ Hz, 1H), 2.68 (dd, $J = 14.6, 5.5$ Hz, 1H), 2.36 (dd, $J =$ 14.8, 4.6 Hz, 1H), 2.32 (dd, J = 14.8, 4.8 Hz, 1H), 1.85 (m, 1H), 1.42−1.21 (m, 3 H), 1.00 (t, J = 7.1 Hz, 3H), 0.91 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, C₆D₆) δ 156.1, 141.9, 128.4, 126.8, 126.2, 62.3, 61.0, 55.8, 53.5, 45.9, 43.6, 40.7, 36.1, 20.3, 14.2, 13.9; HRMS-ESI (M + Na) m/z calcd for C₁₉H₂₇NO₂S₂Na 388.1381, found 388.1377.

(7R,9S)-Ethyl-7-octyl-9-phenyl-1,4-dithia-8-azaspiro[4.5]decane-8-carboxylate 10h and (7R,9R)-Ethyl-7-octyl-9-phenyl-1,4-dithia-8 azaspiro[4.5]decane-8-carboxylate 11h. Yellow oil, starting from 0.8 mmol of 7s, 0.23 g, yield = 65% (entry 8, Table 1) in a ratio of $10h/$ 11h 62/38 in favor of the trans isomer 10h)

10h. ¹H NMR (400 MHz, CDCl₃) δ 7.25−7.12 (m, 5H), 5.20 (dd, J = 9.1, 7.3 Hz, 1H), 4.27 (m, 1H), 4.00 (q, J = 7.[1](#page-2-0) Hz, 2H), 3.28−3.20 $(m, 4H)$, 2.56 (ddd, J = 14.4, 7.3, 1.5 Hz, 1H), 2.50 (dd, J = 14.4, 9.3) Hz, 1H), 2.49 (ddd, $J = 14.1$, 8.8, 1.5 Hz, 1H), 2.09 (dd, $J = 14.1$, 5.5 Hz, 1H), 1.67 (m, 1H), 1.38–1.11 (m, 15 H), 1.04 (t, J = 7.1 Hz, 3H), 0.80 (t, J = 6.8 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 157.0, 143.8, 128.2, 126.6, 126.0, 62.5, 61.4, 55.8, 52.4, 45.9, 44.3, 39.2, 39.1, 38.0, 31.8, 29.5, 29.4, 26.5, 22.7, 14.5, 14.1.

11h. ¹H NMR (400 MHz, CDCl₃) δ 7.25–7.12 (m, 5H), 5.09 (t, J $= 5.1$ Hz, 1H), 4.15 (m, 1H), 3.96 (q, J = 7.1 Hz, 2H), 3.17–3.02 (m, 4H), 2.76 (dd, J = 14.6, 5.1 Hz, 1H), 2.69 (dd, J = 14.6, 5.1 Hz, 1H), 2.36 (dd, $J = 14.8$, 5.1 Hz, 1H), 2.32 (dd, $J = 14.8$, 4.2 Hz, 1H), 1.85 (m, 1H), 1.76−1.01 (m, 15 H), 1.08 (t, $J = 7.1$ Hz, 3H), 0.82 (t, $J =$ 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 128.4, 126.8, 126.2, 62.3, 61.0, 55.8, 53.5, 45.9, 43.6, 40.7, 37.9, 36.1, 32.0, 29.4, 29.6, 26.3, 22.6, 14.6, 14.0; HRMS-ESI (M + Na) m/z calcd for $C_{24}H_{37}NO_2S_2Na$ 458.2163, found 458.2165.

(7S,9S)-Ethyl-7,9-diphenyl-1,4-dithia-8-azaspiro[4.5]decane-8 carboxylate 10j. Yellow oil, starting from 0.6 mmol of 7v, 0.16 g, yield = 65% in a ratio of 10j/11j \geq 96/4 in favor of the *trans* isomer 10j. ¹H NMR (400 MHz, CDCl₃) δ 7.25–7.11 (m, 10H), 5.22 (dd, J = 8.6, 6.3 Hz, 2H), 3.91 (q, J = 7.1 Hz, 2H), 3.31−3.23 (m, 4H), 2.64 (dd, J = 14.4, 8.6 Hz, 2H), 2.52 (dd, J = 14.4, 6.3 Hz, 2H), 0.89 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 157.6, 143.6, 128.2, 126.8, 126.7, 62.4, 61.5, 57.5, 46.0, 39.5, 39.0, 14.1; HRMS-ESI (M + Na) m/ z calcd for $C_{22}H_{25}NO_2S_2Na$ 422.1224, found 422.1225.

(9R,7S)-Ethyl-9-methyl-7-(pyridin-3-yl)-1,4-dithia-8-azaspiro[4.5] decane-8-carboxylate 10k. Yellow oil, yield = 48% in a ratio of $10k/$ 11k 85/15 in favor of the trans isomer 10k. ¹H NMR (400 MHz, CDCl₃) δ 8.46 (m, 1H), 8.42 (d, J = 4.5 Hz, 1H), 7.48 (d, J = 8.0 Hz, 1H), 7.21(m, 1H), 5.13 (t, J = 4.9 Hz, 1H), 4.26 (m, 1H), 4.07−3.93 (m, 2H), 3.30−3.01 (m, 4H), 2.78 (dd, J = 15.0, 4.9 Hz, 1H), 2.69 $(dd, J = 15.0, 4.9 \text{ Hz}, 1\text{H}), 2.44 \text{ (dd, } J = 15.0, 5.1 \text{ Hz}, 1\text{H}), 2.24 \text{ (dd, } J$ $= 15.1, 3.8$ Hz, 1H), 1.42 (d, J = 6.9 Hz, 3H), 1.05 (t, J = 6.9 Hz); ¹³C NMR (101 MHz, CDCl₃) δ 155.0, 146.7, 146.6, 136.6, 122.1, 60.4, 59.8, 52.4, 47.7, 45.3, 44.9, 38.6, 38.0, 19.4, 13.4; HRMS-ESI (M + Na) m/z calcd for $C_{16}H_{22}N_2O_2S_2N_4$ 361.1020, found 361.1022.

(7R,9R)-Ethyl-7-methyl-9-propyl-1,4-dithia-8-azaspiro[4.5] decane-8-carboxylate 10v and (7R,9S)-Ethyl-7-methyl-9-propyl-1,4-dithia-8-azaspiro[4.5]decane-8-carboxylate 11v. .Yellow oil, starting from 0.9 mmol of 7l, 0.20 g, yield = 73% in a ratio of $10v/$ 11v 80/20 in favor of the trans isomer 10v.

10v. ¹H NMR (400 MHz, CDCl₃) δ 4.12 - 3.96 (m, 2H), 3.93 (qdd, J = 7.0, 5.1, 4.8 Hz, 1H), 3.85 (m, 1H), 3.32−3.13 (m, 4H), 2.39 $(dd, J = 14.7, 4.8 Hz, 1H), 2.35 (dd, J = 14.9, 4.8 Hz, 1H), 2.31 (dd, J)$ $= 14.9, 4.3$ Hz, 1H), 2.17 (dd, J = 14.7, 5.1 Hz, 1H), 1.69 (m, 1H), 1.55 (m, 1H), 1.37 (m, 1H), 1.29 (d, J = 7.0 Hz, 3H), 1.25 (m, 1H), 1.18 (t, $J = 7.1$ Hz, 3H), 0.86 (t, $J = 7.3$ Hz, 3H); ¹³C NMR (101) MHz, CDCl3) δ 155.8, 61.7, 60.8, 52.4, 47.9, 46.7, 42.9, 39.4, 39.1, 35.5, 20.4, 19.8, 14.6, 13.8; HRMS $(M + H)^+$ ion by direct probe calcd for $C_{14}H_{26}NO_2S_2$ 304.1405, found 304.1394.

11v. ¹H NMR (400 MHz, CDCl₃) δ 4.34 (m, 1H), 4.26 (m, 1H), 4.09 (q, J = 7.1 Hz, 2H), 3.25–3.13 (m, 4H), 2.45 (ddd, J = 13.9, 9.1, 1.8 Hz, 1H), 2.35 (ddd, $J = 13.6, 8.1, 1.8$ Hz, 1H), 1.98 (dd, $J = 13.9$, 7.5 Hz, 1H), 1.95 (dd, J = 13.6, 7.3 Hz, 1H), 2.18−1.45 (m, 2H), 1.55 $(m, 2H)$, 1.37–1.23 $(m, 2H)$, 1.21 $(d, J = 6.8 \text{ Hz}, 3H)$, 1.98 $(t, J = 7.1 \text{ Hz})$ Hz, 3H), 0.84 (t, J = 7.5 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 156.1, 61.9, 61.2, 52.4, 50.5, 47.6, 45.5, 39.4, 39.0, 35.5, 22.5, 19.6, 14.7, 13.9; HRMS $(M + H)^+$ ion by direct probe calcd for $C_{14}H_{26}NO_2S_2$ 304.1405, found 304.1401.

(7R,9R)-Ethyl-7-methyl-9-octyl-1,4-dithia-8-azaspiro[4.5]decane-8-carboxylate 10t and (7R,9S)-Ethyl-7-methyl-9-octyl-1,4-dithia-8 azaspiro[4.5]decane-8-carboxylate 11t. Yellow oil, starting from 1.0 mmol of 7m, 0.23 g, yield = 62%; entry 5 Table 5, in a ratio of $10t/11t$ 80/20 in favor of the trans isomer 10t.

10t. ¹H NMR (400 MHz, CDCl₃) δ 4.06 (qd, J = 7.1, 3.5 Hz, 2H), 3.92 (qdd, J = 6.8, 5.3, 4.8 Hz, 1H), 3.83 (m, 1[H\)](#page-6-0), 3.32−3.15 (m, 4H), 2.40 (dd, $J = 14.6$, 4.8 Hz, 1H), 2.36 (dd, $J = 15.1$, 5.0 Hz, 1H), 2.30 $(dd, J = 15.1, 4.3 Hz, 1H), 2.17 (dd, J = 14.6, 5.3 Hz, 1H), 1.70 (m,$ 1H), 1.59 (m, 1H), 1.37 (m, 1H), 1.31 (d, J = 6.8 Hz, 3H), 1.16−1.29 (m, 14H), 0.81 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 155.7, 61.6, 60.7, 52.6, 47.9, 46.6, 42.7, 39.3, 39.1, 33.2, 31.8, 29.4, 29.3, 29.2, 26.4, 22.6, 20.3, 14.6, 14.0; HRMS-ESI (M + Na) m/z calcd for $C_{19}H_{35}NO_2S_2Na$ 396.2007, found 396.2005

11t. ¹H NMR (400 MHz, CDCl₃) δ 4.34 (sex, J = 7.1 Hz, 1H), 4.24 (m, 1H), 4.10 (qd, J = 7.1, 4.5 Hz, 2H), 3.24−3.19 (m, 4H), 2.44 $(ddd, J = 13.9, 4.5, 2.0 Hz, 1H), 2.35 (ddd, J = 7.1, 13.9, 2.0 Hz, 1H),$ 1.99 (dd, J = 13.9, 7.1 Hz, 1H), 1.96 (dd, J = 13.9, 7.1 Hz, 1H), 1.99− 1.47 (m, 2H), 1.24−1.17 (m, 18H), 0.84 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 156.1, 62.9, 61.2, 51.0, 47.5, 43.9, 38.9, 38.7, 38.3, 31.8, 29.5, 29.4, 29.2, 26.4, 22.6, 22.5, 14.7, 14.1; HRMS-

ESI $(M + Na)$ m/z calcd for $C_{19}H_{35}NO_2S_2Na$ 396.2007, found 396.2003

(7R,9R)-Ethyl-7-octyl-9-propyl-1,4-dithia-8-azaspiro[4.5]decane-8-carboxylate 10u and (7R,9S)-Ethyl-7-octyl-9-propyl-1,4-dithia-8 azaspiro[4.5]decane-8-carboxylate 11u. Yellow oil, starting from 0.6 mmol of 7 u , 0.16 g, yield = 68%; entry 1 Table 5; in a ratio of $10u/$ 11u 80/20 in favor of the trans isomer 10u.

10u. ¹H NMR (400 MHz, CDCl₃) δ 4.06 (qd, J = 7.1, 2.7 Hz, 2H), 3.74 (m, 2H), 3.30−3.15 (m, 4H), 2.62 (ddd, J [=](#page-6-0) 14.6, 4.8, 1.5 Hz, 2H), 2.19 (dd, J = 14.6, 5.6 Hz, 2H), 1.87−1.73 and 1.58−1.45 (2*m, 4H), 1.37−1.13 (m, 17H), 0.86 (t, J = 7.3 Hz, 3H), 0.86 (t, J = 6.8 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 156.1, 62.9, 60.8, 53.0, 52.7, 44.0, 43.9, 39.1, 39.0, 35.2, 33.0, 31.8, 29.5, 29.4, 29.3, 26.6, 22.6, 19.8, 14.6, 14.1, 13.9; HRMS-ESI (M + Na) m/z calcd for $C_{21}H_{39}NO_2S_2Na$ 424.2320, found 424.2327

11u. ¹H NMR (400 MHz, CDCl₃) δ 4.29–4.19 (m, 2H), 4.09 (q, J $= 7.1$ Hz, 2H), 3.20 (s, 4H), 2.41 (dd, J = 13.9, 8.6 Hz, 2H), 1.94 (dd, J = 13.9, 6.1 Hz, 2H), 1.62−1.56 and 1.49−1.40 (2*m, 4H), 1.32−1.18 $(m, 17H)$, 0.85 $(t, J = 7.6 \text{ Hz}, 3H)$, 0.81 $(t, J = 7.1 \text{ Hz}, 3H)$; ¹³C NMR $(101 \text{ MHz}, \text{CDCl}_3)$ δ 156.4, 63.1, 61.2, 51.4, 51.1, 44.1, 44.0, 40.1, 39.1, 38.6, 37.9, 31.8, 29.5, 29.4, 29.2, 26.4, 22.6, 19.6, 14.7, 14.1, 13.9; HRMS-ESI (M + Na) m/z calcd for $C_{21}H_{39}NO_2S_2Na$ 424.2320, found 424.2323.

Synthesis of (−)-Solenopsine A from 7p. (7R,9R)-Benzyl-7 methyl-9-undecyl-1,4-dioxa-8-azaspiro[4.5]decane-8-carboxylate 8w. To a solution of compound 7p (400 mg, 1.07 mmol, 1 equiv) were added successively trimethyl orthoformate (0.587 mL, 5.36 mmol, 5 equiv), ethylene glycol (0.300 mL, 5.36 mmol 5 equiv), and p-toluenesulphonic acid $(0.04 \text{ g}, 0.21 \text{ mmol } 0.2 \text{ equiv})$. The reaction is followed by TLC, and after 1 h ethyl acetate was added to the crude mixture, which was quenched with a saturated solution of $NAHCO₃$ and extracted twice with ethyl acetate. Then the organic layer was dried over $Na₂SO₄$ and concentrated under vacuum. The crude residue was purified by flash chromatography (eluent, cyclohexane/EtOAc 95:5 to cyclohexane/EtOAc 85:15), to afford trans $8w$ (0,32 g, 67%): ¹H NMR (400 MHz, CDCl₃) δ 7.40–7.27 (m, 5H), 5.17 (d, J = 12.4 Hz, 1H), 5.09 (d, J = 12.4 Hz, 1H), 4.10 (m, 1H), 4.00–3.91 (m, 3H), 3.90−3.81 (m, 2H), 2.14 (dd, J = 14.8, 5.6 Hz, 1H), 2.05 (dd, J = 15.2, 5.6 Hz), 1.97 (dd, J = 14.8, 3.8 Hz, 1H), 1.81 (dd, J = 15.2, 4.1 Hz, 1H), 1.76−1.58 (m, 2H), 1.34 (d, J = 7.1 Hz, 3H), 1.37−1.22 (m, 16H), 0.88 (t, J = 6.8 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 155.6, 137.1, 128.6, 128.0, 106.7, 66.9, 64.0, 63.9, 51.5, 46.6, 39.6, 35.9, 34.2, 32.1, 29.8, 29.7, 29.5, 26.8, 21.0, 14.3; HRMS-ESI (M + Na) m/z calcd for $C_{27}H_{44}NO_4$ 446.3270 found 446.3261.

(7R,9R)-7-Methyl-9-undecyl-1,4-dithia-8-azaspiro[4.5]decane 10w. To a solution of 8w (300 mg, 0.64 mmol, 1 equiv) in dry CH_2Cl_2 (5 mL) were added successively 1,2-ethanedithiol (0.268 mL, 3.2 mmol, 5 equiv) and BF_3Et_2O (0.406 mL, 3.2 mmol, 5 equiv) dropwise at 0 °C. After 24 h at room temperature, the solution was quenched with a solution of NaOH at 0 °C and extracted three times with CH_2Cl_2 . Then the organic layer was dried and concentrated under vacuum before purification by flash chromatography (eluent, cyclohexane to cyclohexane/EtOAc 70:30) to give trans 10w (0,175 g, 76%) as a yellow oil: ¹H NMR (400 MHz, CDCl₃) δ 3.32−3.27 (m, 4H), 3.20 (m, 1H), 3.02 (m, 1H), 2.26 (dd, J = 13.4, 3.9 Hz, 1H), 2.22 $(dd, J = 13.1, 2.6 Hz)$, 1.92 (dd, J = 13.4, 5.6 Hz, 1H), 1.82 (dd, J = 13.1, 6.7 Hz, 1H), 1.57−1.51 (m, 2H), 1.35−1.20 (m, 16H), 1.16 (d, J $= 6.7$ Hz, 3H), 0.88 (t, J = 6.8 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 65.3, 51.6, 49.2, 46.2, 46.0, 39.1, 38.7, 34.6, 32.1, 29.8, 29.7, 29.5, 26.9, 22.8, 21.0, 14.3; HRMS-ESI (M + Na) m/z calcd for $C_{19}H_{37}NNaS_2$ 366,2265 found 366,2257.

(2R,6R)-tert-Butyl-2-methyl-6-undecylpiperidine-1-carboxylate 12. To a stirred solution of dithioketal 10w (0,140 g, 0.44 mmol, 1 equiv) in THF (5 mL) were added successively di-tert-butyl dicarbonate (0.191 g, 0.88 mmol, 2 equiv) and DMAP (25 mg, 0.02 mmol, 0.05 equiv) at 0 °C. After 1 h at room temperature the resulting solution mixture was washed with a solution of $NH₄Cl$ and extracted with ethyl acetate. The organic layer was dried over $Na₂SO₄$ and concentrated under vacuum. The resulting oil was directly engaged in the following steps without further purification. To a solution of the crude oil in ethanol (5 mL) was added freshly prepared W_2 Raney nickel (ca. 1 g). The resulting suspension was heated at reflux for 2 h and then cooled to room temperature. The suspension was then filtered through Celite, and the filtrate was concentrated under reduced pressure. The residue was dissolved in 1 M aqueous NaOH and extracted with dichloromethane. The combined organic extracts were washed with brine and dried over $Na₂SO₄$. Evaporation of the solvent followed by column chromatography (eluent, cyclohexane to cyclohexane/EtOAc 80:20) gave compound 12 (100 mg, 70%) as a colorless oil. Spectral data are identical with those reported.⁴⁸

(−)-Solenopsine A. Trifluoroacetic acid (1 mL) was added to a solution of 12 (80 mg, 0.22 mmol) in CH_2Cl_2 (1 mL)[, a](#page-15-0)nd the reaction mixture was stirred at room temperature for 2 h. The mixture was evaporated, and the residue was basified with 2 N NaOH. The solution was extracted with CH_2Cl_2 three times. The extracts were dried with $Na₂SO₄$ and evaporated. The residue was purified with chromatography (eluent, CHCl₃/MeOH 5:1) to yield (−)-solenop-
sine A (50 mg, 87%) as an oil. [α]_D = −1.21 (*c* 0.94, CH₃OH), lit.⁴⁹ $[\alpha]_{\text{D}} = -1.30$ (c 1.30, CH₃OH). Spectral data are identical with those reported. $^\mathrm{48}$

Synthesis of 241D from 7o. (2R,6S)-Benzyl-4,4-dimethoxy-[2](#page-15-0) methyl-[6-n](#page-15-0)onylpiperidine-1-carboxylate 9′x. To compound 7o (0.25 g, 0.67 mmol, 1 equiv) were added successively trimethyl orthoformate (0.367 mL, 3.35 mmol, 5 equiv) and p-toluenesulphonic acid (127 mg, 0.67 mmol, 1 equiv). After 1 h, ethyl acetate was added to the crude mixture, which was quenched with a saturated solution of $NAHCO₃$ and extracted twice with ethyl acetate. Then the organic layer was dried and concentrated under vacuum. The crude oil mixture was separated by flash chromatography (eluent, cyclohexane/EtOAc 95:5 to cyclohexane/EtOAc 85:15) to afford pure cis isomer 9′x as an oil (172 mg, 61%): ¹H NMR (400 MHz, CDCl₃) δ 7.28–7.19 (m, 5H), 5.09 (d, J = 12.6 Hz, 1H), 5.05 (d, J = 12.6 Hz, 1H), 4.37 (quintd, J = 7.1, 3.7 Hz, 1H), 4.21 (qd, J = 7.1, 2.1 Hz, 1H), 3.11 (s, 6H), 1.87 (dt, $J = 14.0, 2.1$ Hz, 1H), 1.81 (ddd, $J = 13.7, 3.7, 2.1$ Hz, 1H), 1.73 (dd, J $= 14.0, 7.2$ Hz, 1H), 1.69 (dd, $J = 13.7, 7.7$ Hz, 1H), 1.61 (m, 2H), 1.23 (d, J = 7.1 Hz, 3H), 1.22–1.17 (m, 12H), 0.81 (t, J = 6.8 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 155.6, 137.0, 128.7, 128.4, 127.9, 98.7, 67.0, 50.7, 47.9, 47.4, 46.3, 36.2, 35.7, 33.6, 31.9, 29.7, 29.6, 29.3, 27.2, 22.7, 21.5, 14.1; HRMS-ESI (M + Na) m/z calcd for C₂₅H₄₁NO₄Na 442.2933 found 442.2948.

(2R,6S)-Benzyl-2-methyl-6-nonyl-4-oxopiperidine-1-carboxylate **13h.** To a solution of $9'x$ (0.150 g, 0.35 mmol, 1 equiv) in CH_2Cl_2 (0.5 mL) was added slowly TFA/H₂O $(1:1, 0.5 \text{ mL})$ at room temperature. After 1 h the mixture was quenched with NaOH (1 M) and extracted twice with CH_2Cl_2 . Then the organic layer was dried and concentrated under vacuum. The yellow oil obtained was filtered through a pad of silica and washed with ethyl acetate to furnish after evaporation of the solvent compound 13 as a yellow oil in a quantitative yield: ¹H NMR (400 MHz, CDCl₃) δ 7.31–7.21 (m, 5H), 5.12 (d, $J = 12.4$ Hz, 1H), 5.08 (d, $J = 12.4$ Hz, 1H), 4.72 (m, 1H), 4.59 (m, 1H), 2.65 (dd, J = 14.9, 7.7 Hz, 1H), 2.60 (d, J = 14.6, 7.5 Hz, 1H), 2.27 (ddd, J = 14.9, 3.7, 1.6 Hz, 1H), 2.23 (ddd, J = 14.6, 4.2, 1.6 Hz, 1H), 1.59−1.35 (m, 2H), 1.22 (d, J = 6.9 Hz, 3H), 1.26−1.10 (m, 12H), 0.81 (t, J = 6.9 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 208.1, 155.6, 136.5, 128.5, 128.1, 128.0, 67.5, 53.1, 48.8, 45.5, 43.7, 36.9, 31.9, 29.5, 29.3, 29.3, 26.3, 22.7, 21.5, 14.1; HRMS-ESI (M + Na) m/z calcd for C23H35NO3Na 396.2515 found 396.2511

(2R,4S,6S)-2-Methyl-6-nonylpiperidin-4-ol: (+)-Alkaloid 241D. To a solution of 13 (0.135 g, 0.35 mmol 1 equiv) in MeOH (5 mL) was added Pd/C (5%, 15 mg) under H_2 atmosphere (1 atm). After 24 h, the solution was filtered through a pad of Celite and washed 3 times with MeOH. After concentration under vacuum, to the crude oil product was added slowly NaBH4 (13 mg, 0.35 mmol, 1 equiv) at 0 °C. After 15 min at room temperature, the solution was quenched with a solution of brine and concentrated under vacuum. The residue was then diluted with ethyl acetate and washed with $H₂O$. Then the organic layer was dried and concentrated under vacuum. The yellow oil obtained was purified by flash chromatography (eluent, EtOAc to EtOAc/MeOH 90:10) to give (+)-alkaloid 241D (71 mg, 84% over two steps) as colorless needles: mp 107−108 °C; [α]_D = +5.66 (α 0.60

MeOH, 95% ee), lit.¹⁷ $[\alpha]_{\text{D}}$ = +5.90 (c 0.65 MeOH, \geq 99% ee). ¹H NMR (400 MHz, CDCl₃) δ 3.66 (tt, J = 5.0, 11.2 Hz, 1H), 2.69 (m, 1H), 2.55 (m, 1H), 1.96 (dt, J = 11.2, 5.0, 2H), 1.65−1.31 (m, 3H), 1.30−1.13 (m, 1H), 0.97 (q, J = 11.2 Hz, 1H), 0.91 (q, J = 11.2 Hz, 1H), 0.87 (t, $J = 7.0$ Hz, 1H).

■ ASSOCIATED CONTENT

6 Supporting Information

¹H NMR and ¹³C NMR of 5 to 11 are described and copies of spectra are given. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

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